

## REPORT DOCUMENTATION PAGE

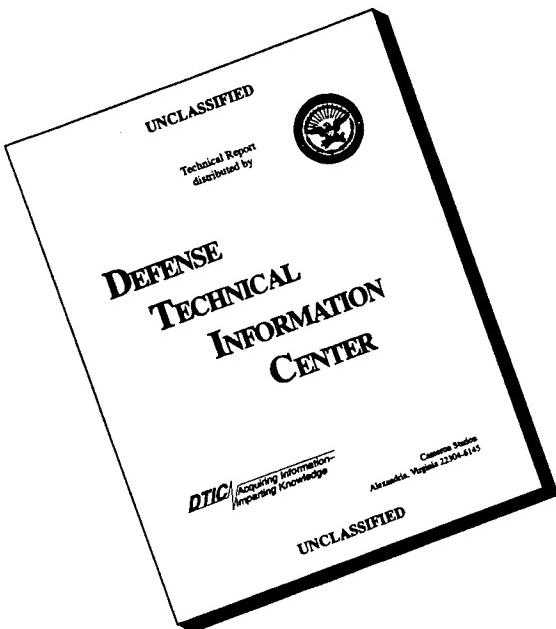
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13. ABSTRACT (Maximum 200 words)  METHODOLOGY AND RESULTS OF THE ANALYSIS OF 102 DENVER GROUND WATER SAMPLES FOR VARIOUS TARGET COMPOUNDS ARE PRESENTED. THE OBJECTIVE OF THE DENVER SOIL AND GROUND WATER PROJECT IS TO PREVENT FURTHER ENVIRONMENTAL CONTAMINATION, CONTROL/CONTAIN EXISTING CONTAMINATION, AND UNDERTAKE MITIGATIVE EFFORT AS DEEMED NECESSARY. AS PART OF THE PROBLEM DEFINITION PHASE OF THIS PROJECT A LARGE NUMBER OF DENVER WATER SAMPLES WERE COLLECTED AND ANALYZED AT BSRC. THE PURPOSE OF THIS SURVEY WAS THE CONFIRMATION OF SEMI-QUANTIFICATION OF SPECIFIC TARGET COMPOUNDS KNOWN OR SUSPECTED TO BE PRESENT IN THE WATER SAMPLES. GB/MS WAS THE METHOD OF CHOICE FOR THIS WORK; THE OVERALL METHODOLOGY BEING SIMILAR TO THAT EMPLOYED FOR PRIORITY POLLUTANT ANALYSIS OF WATER. ALSO DESIRED WAS IDENTIFICATION AND SEMIQUANTIFICATION OF ANY OTHER SIGNIFICANT COMPONENTS. IN ADDITION, PH, CONDUCTANCE, CHLORIDE ION, SULFATE ION, AND IONIC COPPER WERE DETERMINED.		
DATA QUALITY INDEXED 1		
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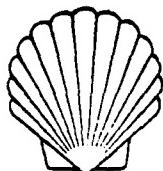
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## SHELL CHEMICAL COMPANY

A DIVISION OF SHELL OIL COMPANY

P. O. BOX 2171

DENVER, COLORADO 80201

April 16, 1981

Colorado Department of Health  
Water Quality Control Division  
ATTN Mr. Robert Shukle, Industrial  
Waste Consultant  
4210 East 11th Avenue  
Denver, CO 80220

Gentlemen:

Shell has conducted a survey of the groundwater on its leasehold on the Rocky Mountain Arsenal to assess groundwater quality. The analytical results from the survey as they became available have been reviewed with members of the Technical Review Committee at the last several meetings. These results have been summarized in a Technical Progress Report (M-9-81) which was recently issued by Shell Development Company. A copy of this report and a map showing well locations are attached for your information.

Yours very truly,

J. H. Knaus  
Plant Manager

### Attachments

cc: Commander  
Rocky Mountain Arsenal  
Commerce City, CO 80022

Commander  
Rocky Mountain Arsenal  
ATTN SARRM-TOE-C, Greg Ward, Bldg. 741  
Commerce City, CO 80022

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ANALYSIS OF DENVER WATER SAMPLES  
TECHNICAL PROGRESS REPORT NO. M-9-81

By E. J. Silveira

ANALYSIS OF DENVER WATER SAMPLES



Project No. 83596.00

Denver Plant Ground Water Program

Technical Progress Report No. M-9-81

Date of Issue:

Written and Reviewed : E. J. Silveira

Participants : G. W. Campbell, D. R. Farinha,  
K. S. Williams, B. Calog

Approval : R. A. Newman

References : MOLR 2383, 2606

SHARED - Under the Research Agreement between SIRM,  
and Shell Oil Company dated January 1, 1960,  
as amended.

ABSTRACT

Methodology and results of the analysis of 102 Denver groundwater, samples for various target compounds are presented.

LHINK Keywords: Water, analysis, Denver, groundwater, Rocky Mountain, arsenal, Colorado, VOA, pH, conductance, chloride, ion, sulfate, copper, method, run-off, well water, effluent

## ANALYSIS OF DENVER WATER SAMPLES

### Introduction

The objective of the Denver Soil and Groundwater Project is to prevent further environmental contamination, control/contain existing contamination, and undertake mitigative effort as deemed necessary. As part of the Problem Definition Phase of this project a large number of Denver water samples were collected and analyzed at BSRC. The purpose of this survey was the confirmation and semi-quantification of specific target compounds known or suspected to be present in the water samples. GC/MS was the method of choice for this work; the overall methodology being similar to that employed for priority pollutant analysis of water. The list of desired target compounds is shown in Table 1. Also desired was identification and semi-quantification of any other significant components.

In addition, pH, conductance, chloride ion, sulfate ion and ionic copper were determined.

### Sampling and Storage

The sampling protocol for the Denver Plant Groundwater Survey Program is as follows.

#### Preparation of the Wells Before Sampling Water

Wells are to be drilled at designated spots on the properties, lined with polyvinyl chloride pipe, and capped. Each well is to be numbered by stencil "Denver Plant Groundwater Well No. \_\_\_\_" (1 through n). Each well is to have an altitude benchmark referenced to the U.S.G.S. benchmark. Prior to each sampling of the well water, the height of the water will be determined, measured from the top of the well, and referenced to the benchmark. By use of a submersible pump (pumps which require priming or vacuum are unsuitable) and a length of Teflon tubing, about 10 volumes of water (as compared to the amount in the well casing) will be pumped out of the well. If recovery times are short and the pumping operation can be accomplished in a reasonable time frame, pumping will be done before each sampling. After each pumping operation, Denver drinking water will be used to rinse the pump. The Teflon tube should require no rinsing but may require drying by use of a piece of gauze or similar clean material.

#### Sampling the Water

In advance of the sampling operation, new quart bottles will be rinsed with methylene chloride and dried at 100°C. Smaller bottles or vials, 40-ml, will be supplied by Westhollow Research Center (WRC)(G. W. Stanko). All bottle caps will contain Teflon liners also to be supplied by WRC. Each bottle will be carefully stored and handled so as to avoid contamination.

The actual water sample will be obtained by bailing as follows: Immediately after pumping, the sample bottle will be placed into a weighted, stainless cage. By use of a stainless steel chain, the bottle will be quickly lowered to the bottom of the well and allowed to fill with water. The full bottle will be brought to the surface, capped, labeled with tape labels, and placed in an

ice bath. The 40-ml size bottle will be sampled in a similar manner, with care being taken to insure liquid full bottles (no air space) upon insertion of the Teflon septum. Four of the 40-ml bottles will be placed in a wide-mouth quart jar and cushioned with wads of paper. After capping, the quart jar will be placed in an ice bath. Two quart and three 40-ml samples of water shall be obtained from each well at each sampling. The full bottles will be returned to the laboratory and placed in a clean refrigerator at a temperature of 0 to 5°C (32 to 41°F); freezing must be avoided. The refrigerator will be used for no other purpose during this groundwater survey program. After the completion of the sampling for the period, or when about 20 wells have been sampled, the samples will be sent to Modesto (BSRC) by refrigerated truck or by air freight in a manner which will keep the bottles between 0 and 5°C. Upon receipt of the samples at BSRC, they will be placed in a refrigerator (0 to 5°C) pending analysis.<sup>a)</sup>

Sampling of the wells at Denver will be coordinated with the work load at BSRC to insure volatile analyses within two weeks of sampling. Extraction should be accomplished within one to two weeks of sampling. Extracts may be sealed in airtight vials and stored in a freezer to await analysis.

#### Records

A log book will be maintained for recording all sampling data. Among the entries will be the names of persons doing the sampling, weather data, height of the water in the well, any variance in the sampling procedure, and unusual events. The log book will be signed by the responsible person after each series of sampling.

A map showing the location of each well is available on request from the Denver Plant.

#### Instrumentation and Methodology for Volatile Organics Analysis (VOA)

The general scheme for VOA analysis is shown in flowchart form in Figure 1. Specific analytical conditions are given in Table 2.

##### VOA Subsampling Procedure

The VOA vial was removed from the refrigerator and allowed to come up to room temperature. The vial was opened and the tip of a 1-ml syringe (no needle attached) was inserted well below the surface and a 1.0-ml sample was drawn very slowly into the syringe. The internal standard was then added by syringe directly into the subsample. A needle was then immediately attached and the sample was injected into the sparging vessel through the septum valve.

##### VOA Standards

VOA reference standard solutions were freshly prepared at least once each week. The reference standard contained 20 µg/ml each of the internal standard and the target compounds in methanol solution. At least one reference standard run was made each day that samples were run. The reference standard run sample was prepared by syringe addition of 5 µl of reference standard solution to 1.0-ml of distilled water previously boiled to remove volatiles. The purpose of this reference standard run was to (1) ensure the reliability of the analysis, and (2) to update response

a) The procedure for analysis is described separately.

factors upon which the quantification was based. A typical VOA reference standard run is shown in Figure 2.

VOA internal standard solutions were freshly prepared at least once each week. This solution contained 20  $\mu\text{g}/\text{ml}$  of only the internal standard, bromochloromethane, in methanol. A 5  $\mu\text{l}$  aliquot of the internal standard solution (100 ng bromochloromethane) was added by syringe to each sample and blank immediately prior to introduction to the sparging vessel. Blank runs, containing only internal standard added to distilled, previously boiled water, were made at least once each day.

VOA reference standard and internal standard solutions were kept refrigerated when not in use.

A typical gas chromatogram/mass spectrum (GC/MS) of the VOA is shown in Figure 2.

#### Extraction Procedure

Filtration - Prior to extraction samples were filtered through a 0.45  $\mu\text{m}$  Millipore membrane filter, catalogue number HAWP-047-00, covered with a Millipore glass fiber prefilter, catalogue number AP25-047-00. A Millipore filter holder with glass frit was used.

#### pH Adjustment

Prior to pH adjustment and extraction approximately 100-ml of each sample was retained and submitted for pH, conductance, chloride ion, sulfate ion and ionic copper determinations.

The volume of the remaining filtrate was noted and the filtrate transferred to a 2-l separatory funnel. Well samples 1A-46A were subjected to extraction without pH adjustment.

The pH of all other samples were adjusted to pH 11 with sodium hydroxide pellets. After extraction the aqueous phase was then adjusted to pH 2 with 6N HCl and reextracted.

#### Extraction

The aqueous solutions were extracted with 120-ml  $\text{CH}_2\text{Cl}_2$  and the  $\text{CH}_2\text{Cl}_2$  phase collected. The aqueous solutions were then extracted twice more with 80-ml portions of fresh  $\text{CH}_2\text{Cl}_2$  and the three extracts combined.

#### Concentration

The combined  $\text{CH}_2\text{Cl}_2$  extracts were concentrated as rapidly as possible in a Kuderna-Danish Evaporator equipped with a 3-ball Snyder column to a volume of approximately 10-ml. The extracts were then further concentrated to approximately 0.8-ml by blowdown with dry nitrogen. An aliquot of a standard solution of  $d_{10}$ -anthracene internal standard (containing 100  $\mu\text{g}$ ) was then added to the extract concentrates and the volume adjusted to 1.0-ml. The extracts derived from the basic solutions are the base-neutral (BN) extracts and those derived from the acidic solutions are the acid extractables (AE). Those extracts derived from solutions without pH adjustment are listed in the result table under the BN heading.

## Instrumentation and Methodology for GC/MS Analysis for Extractable Organics

The general scheme for extractable organics analysis is shown in flowchart form in Figure 3. Specific analytical conditions are given in Table 3.

### BN Standards

BN reference standard solutions were freshly prepared at least every 3-4 weeks. The reference standard contained 100 µg/ml each of the internal standard and the target compounds in methylene chloride solution. At least one reference standard run was made each day that samples were run. The purpose of this reference standard run was (1) to ensure the reliability of the analysis, and (2) to update response factors upon which the quantification was based. A typical reference standard run is shown in Figure 4.

BN internal standard solutions were freshly prepared at least every 3-4 weeks. This solution contained 1 mg/ml of only the internal standard, d<sub>10</sub>-anthracene, in methylene chloride. A 100 µl aliquot of the internal standard solution (100 µg d<sub>10</sub>-anthracene) was added by syringe to each sample extract after nitrogen blowdown.

BN reference standard and internal standard solutions were kept refrigerated when not in use.

A typical GC/MS of a BN extract is shown in Figure 4.

Samples 1A-4A, 9A, 13A, 16A, 17A, 23A-25A, 32A-34A, and 39A-41A were run on the OV-101 packed column (conditions A in Table 3). Samples 5A-8A, 10A-12A, 14A, 15A, 18A-22A, 26A-31A, 35A-38A, 42A-46A, 16B, 24B, 40B, were run on the glass capillary column (conditions B in Table 3). All other samples were run on the fused silica capillary column (conditions C in Table 3).

### Data Processing of GC/MS Runs

The general scheme for data processing is shown in flowchart form in Figure 5. Details of the data processing procedures and programs are to be documented in a separate report.

### pH Measurement

The pH of the water samples was determined by using a Corning 110 Digital pH Meter with a Metrohm EA 121 combination glass electrode.

### Conductance Measurement

Conductance was determined using a Western Instruments Model 210 Conductivity Meter with a Model 219-200 flow through cell. The calibration solution was 0.01 M KCl.

### Chloride Ion Determination

Chloride ion was determined by silver nitrate titration using a Metrohm E 436 Potentiograph with a Metrohm EA 246 silver electrode.

### Sulfate Ion Determination

Sulfate ion was determined turbidimetrically as BaSO<sub>4</sub> using a Hach Model 2100 Turbidimeter.

### Ionic Copper Determination

Ionic copper was determined using a colorimetric method - EMS 2C14/59.

### Results

The analytical results are presented in Table 4. Concentration values (in ug/l or ppb) are given for target compounds exceeding the nominal minimum detectable concentration of 10 ppb. Concentration values for other detected peaks, including tri- and tetrachlorethylene, DMK (acetone), CH<sub>2</sub>Cl<sub>2</sub> and THF, are given if the peak could be tentatively identified and the concentration exceeded 200 ppb.

FIGURE 1

VOA SCHEME

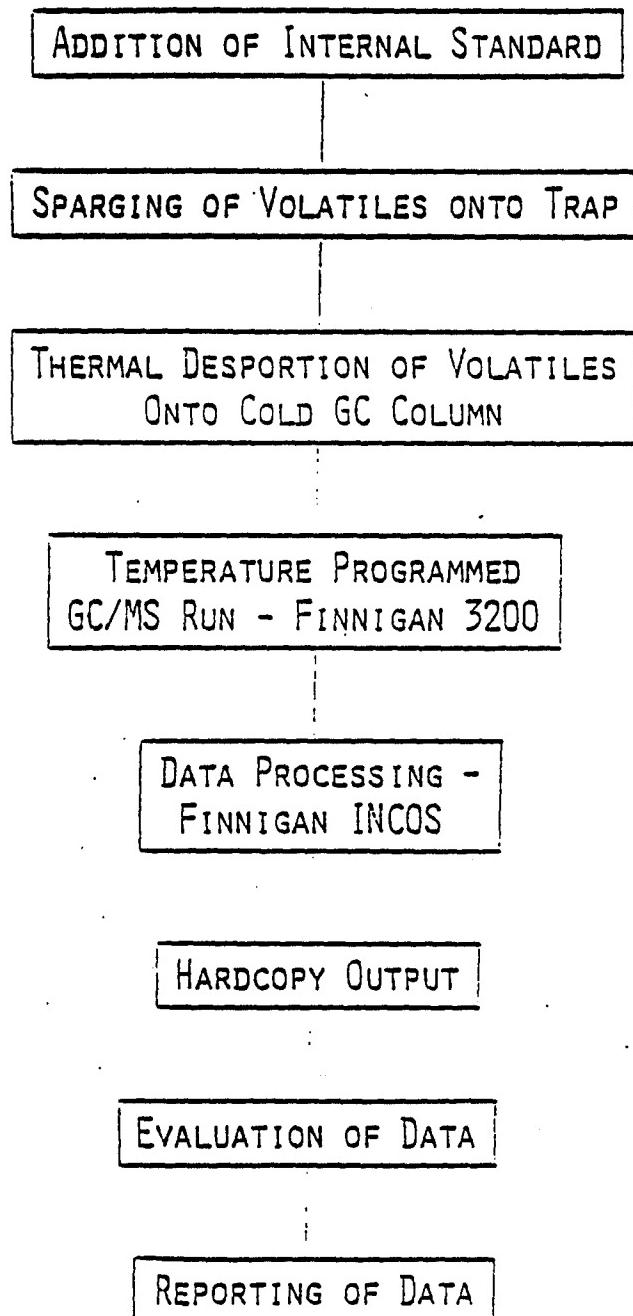


FIGURE 2  
VOA STANDARDS

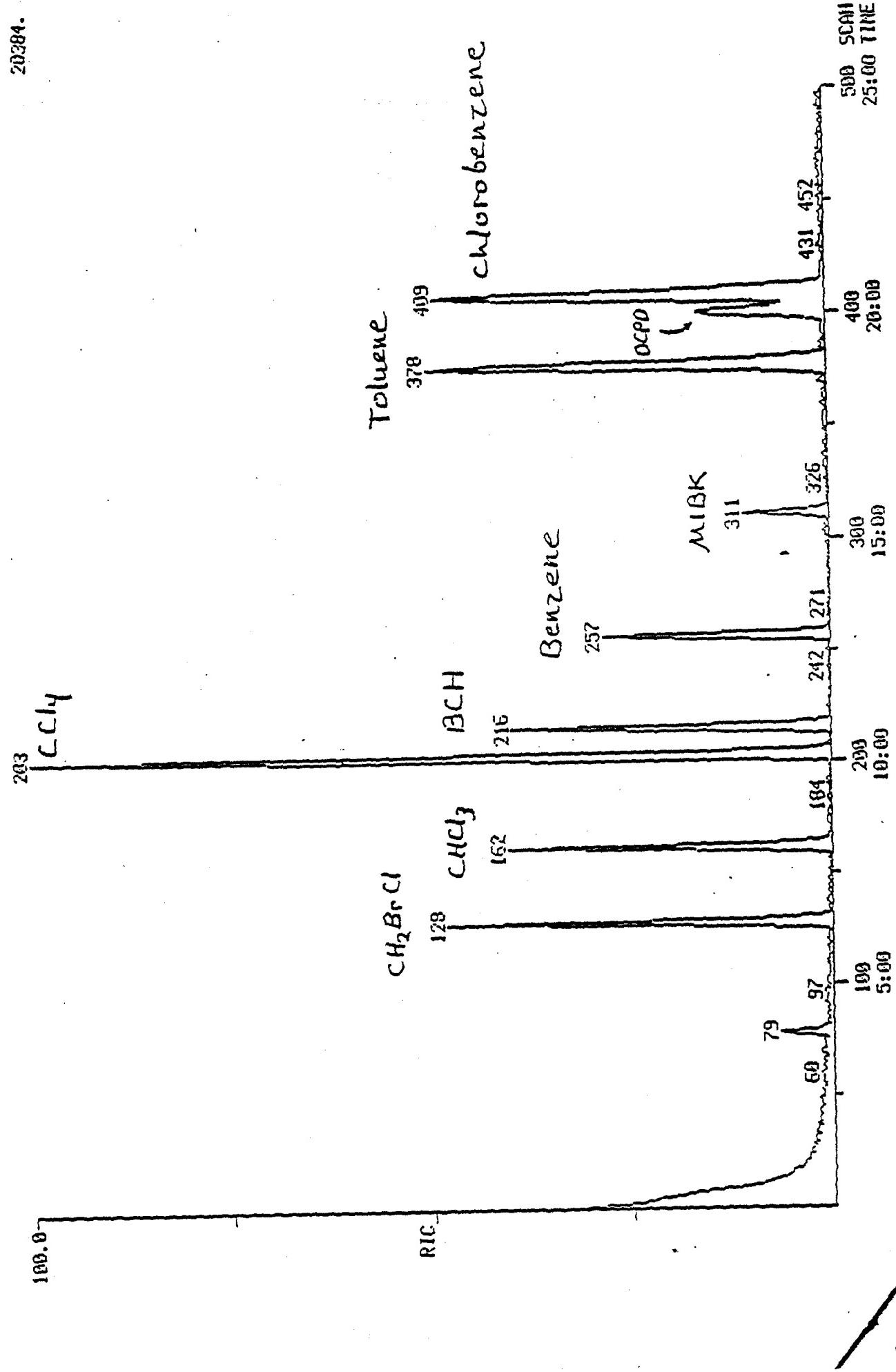
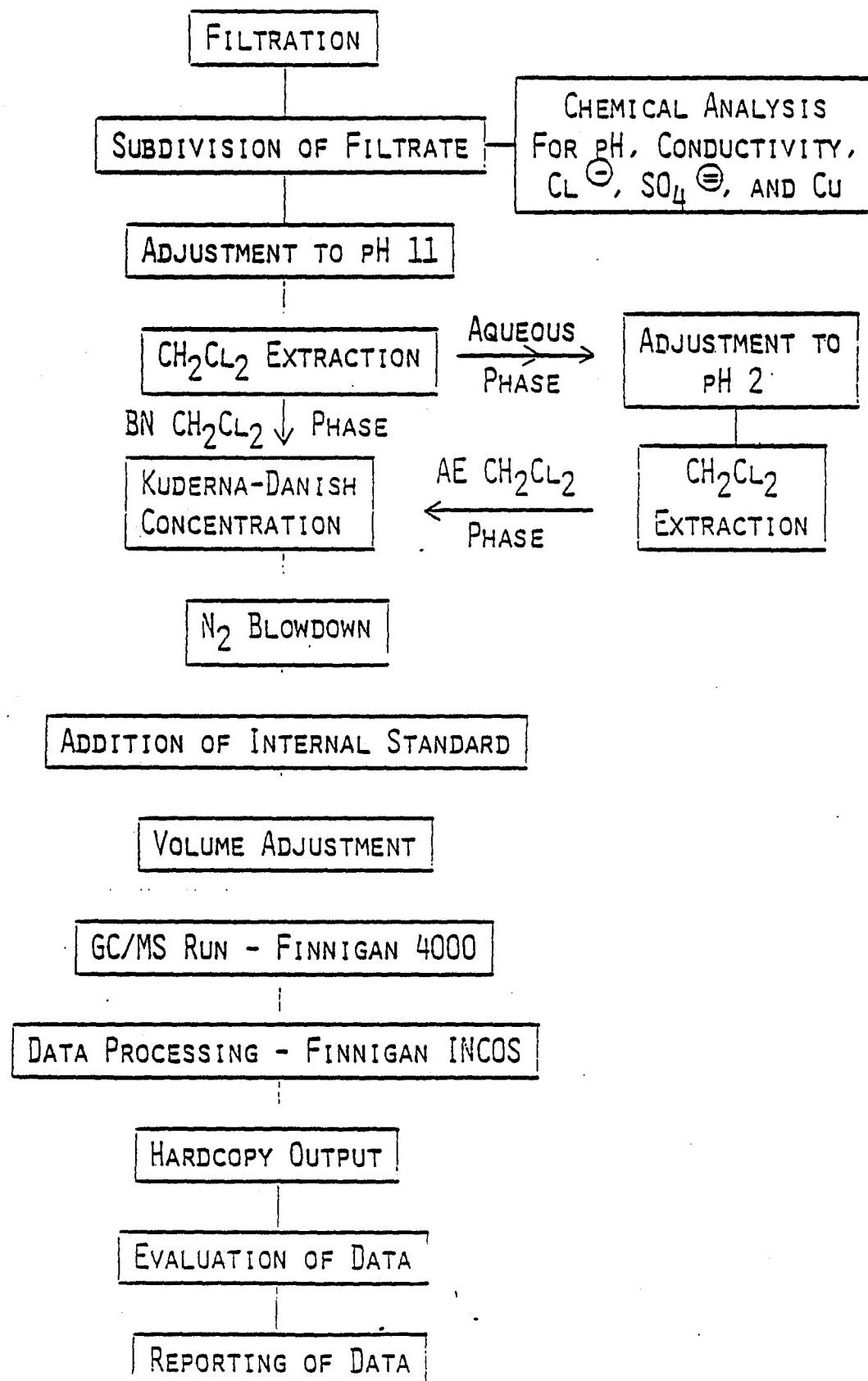


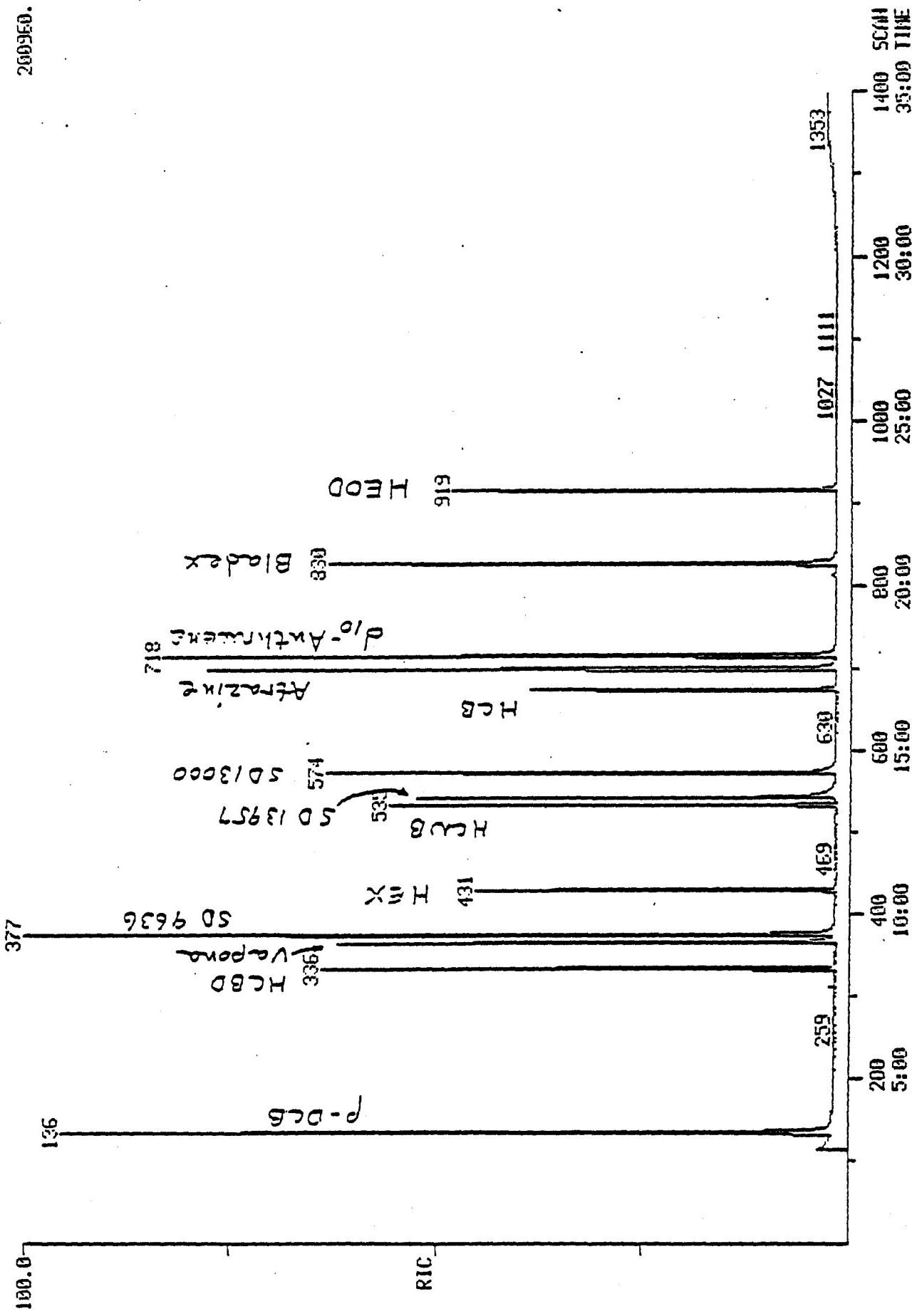
FIGURE 3

LLE SCHEME



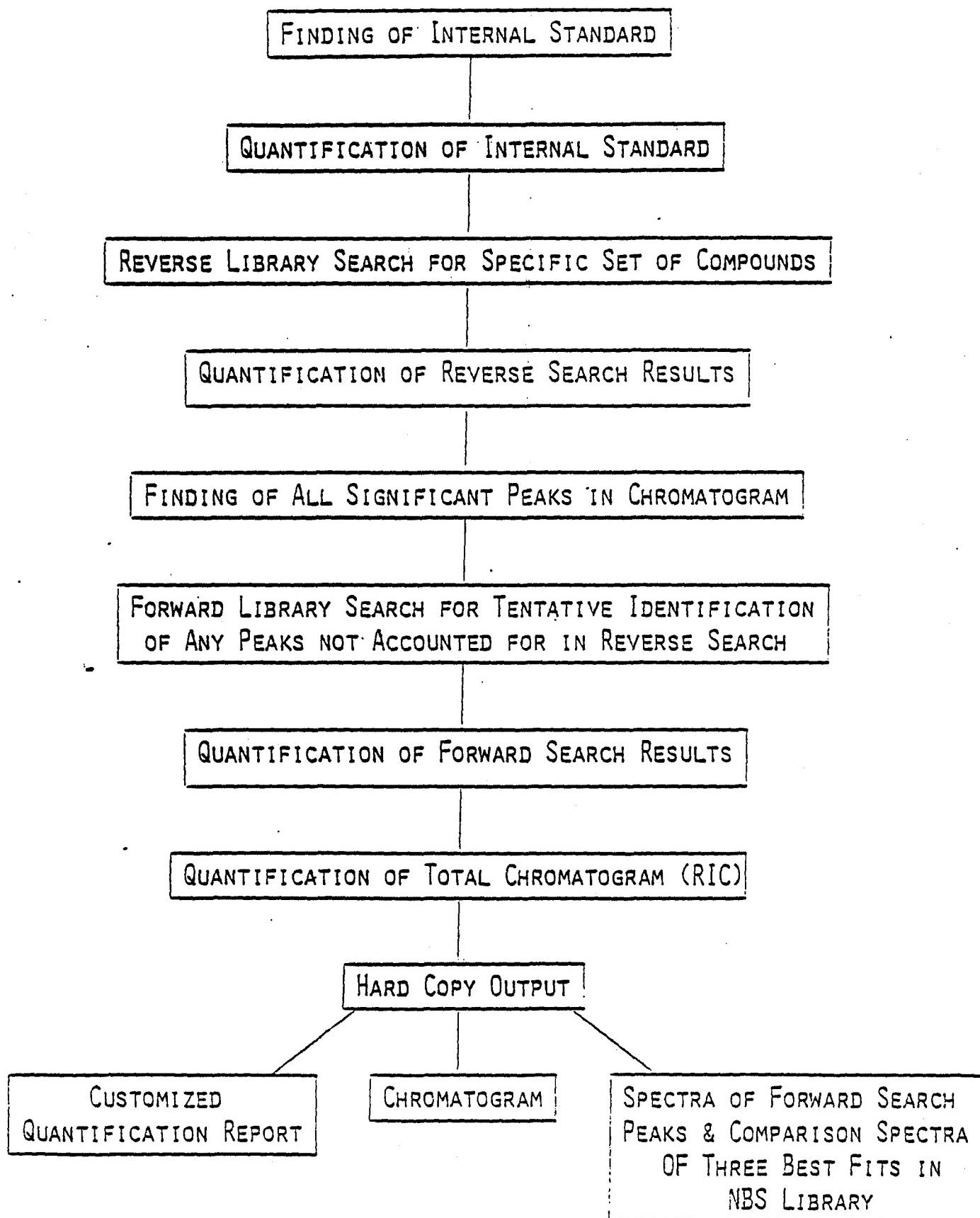
FIGURE

BN STANDARDS



10

FIGURE 5  
DATA PROCESSING SCHEME



DENVER PLANT GROUNDWATER ANALYSIS

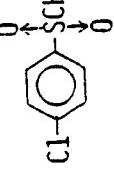
VOLATILE ORGANICS	<u>M<sub>w</sub></u>	QUANTIFICATION ION	ASSIGNMENT		RELATIVE ABUNDANCE
1. CHLOROFORM	118	83	$M^+ - Cl^-$		100
2. CARBON TETRACHLORIDE	152	117	$M^+ - Cl^-$		100
3. BICYCLOHEPTADIENE (BCH)	92	91	$M^+ - H^-$		100
4. BENZENE	78	78	$M^+$		100
5. METHYLISOBUTYLKETONE (MIBK)	100	100	$M^+$		
6. TOLUENE	92	91	$M^+ - H^-$		100
7. DICYCLOPENTADIENE (DCPD)	132	66			100
8. CHLOROBENZENE	112		$M^+$		100



8. CHLOROBENZENE

DENVER PLANT GROUNDWATER ANALYSIS

EXTRACTABLE ORGANICS	<u>MW</u>	QUANTIFICATION ION	ASSIGNMENT	RELATIVE ABUNDANCE	
				$\text{M}^+$	$\text{M}^+ - \text{Br}$ ISOTOPE
1. DICHLOROBENZENES	146	146	$\text{M}^+$	100	
2. DIISOPROPYL METHYLPHOSPHONATE (DIMP)	180	97	$\text{HO}-\text{P}(=\text{O})-\text{OH}^+$	100	
(CH <sub>3</sub> ) <sub>2</sub> CHO-P(=O)CH <sub>3</sub>					
3. DIBROMOCHLOROPROPANE (DBCP)	234	157	$\text{M}^+ - \text{Br}$ ISOTOPE	100	
CH <sub>2</sub> BrClBrCH <sub>2</sub> Cl					
4. HEXACHLOROBUTADIENE	258	225	$\text{M}^+ - \text{Cl}$ ISOTOPE	100	
5. VAPONA	220	109	(CH <sub>3</sub> O) <sub>2</sub> P(=O)O <sup>+</sup>		
CH <sub>3</sub> O-P(=O)CH <sub>3</sub>					
6. SD 9636	158	158	$\text{M}^+$	100	
Cl-C <sub>6</sub> H <sub>4</sub> -SCl <sub>3</sub>					
7. HEXACHLOROCYCLOPENTADIENE (HEX)	270	237	$\text{M}^+ - \text{Cl}$ ISOTOPE		
8. HEXACHLORONORBORNADIENE (HCNB)	296	263	$\text{M}^+ - \text{Cl}$ ISOTOPE	100	
9. SD 13957	174	159	$\text{M}^+ - \text{CH}_3$	100	
Cl-C <sub>6</sub> H <sub>4</sub> -SCH <sub>3</sub>					

	RELATIVE ABUNDANCE	ASSIGNMENT
	QUANTIFICATION ION	
10. SD 13000	190	 Cl 
	111	Cl <sub>2</sub> 
11. HEXACHLOROBENZENE (HCB)	282	100
12. ATRAZINE	215	M <sup>+</sup> - CH <sub>3</sub> 65
13. BLADEX	240	M <sup>+</sup> - ET (H TRANSFER)      45
	212	
14. DIELDRIN (HEOD)	356	C <sub>6</sub> H <sub>7</sub> <sup>7+</sup> 100
	79	

OTHER COMPOUNDS WHICH HAVE BEEN DETERMINED IN SELECTED SAMPLES BY  
SPECIFIC RESIDUE METHODS:

- CYTIC RESIDUE IN INSECT**



TABLE 2 - Conditions for Volatile Organics Analysis

Instrument: Finnigan 3200 GC/MS equipped with a Chemical Data Systems Model 310 Concentrator

Trapping Column: 2' x 1/8" OD stainless steel packed with Tenax 60/80 mesh

Sparge Gas: Helium

Sparge Rate: 30 ml/minute

Sparge Cycle: 12 minutes with sparging vessel at 50°C onto trapping column at room temperature

Desorb Cycle: 4 minutes at 180°C onto GC column at room temperature

Trap Bakeout Cycle: 4 minutes at 280°C

Sample Size: 1 ml or 5 ml

GC Column: 10' x 2mm ID glass column packed with 1% SP-1000 on CarboPack B, 60/80 mesh

Temperature Program: 50°C for 4 minutes, then programmed to 200°C at 10°/minute, then held at 200°C for 13.5 minutes

Injector Temperature: 200°C

Carrier: Helium at 30 ml/minute

Separator: Glass jet

Separator Temperature: 200°C

Transfer Line Temperature: 200°C

MS Mode: Electron Impact (EI)

Electron Energy: 70 eV

Electron Multiplier Voltage: Approximately 1700 V

Emission Current: 1.0 ma

Preamplifier Sensitivity: 10<sup>-7</sup> a/v

Scan Range: 40-540 amu

Scan Time: 3 seconds

Scans per Run: 650

Internal Standard: 100 ng bromochloromethane

TABLE 3 - Conditions for Extractable Organics Analysis

Instrument: Finnigan 4000 GC/MS

GC Columns: (A) 6' x 2mm ID glass column packed with 3% OV-101 on  
Gas Chrom Q, 80/100 mesh  
(B) 30m x 0.25mm ID glass WCOT capillary column coated with SE-54  
(C) 30m x 0.25mm ID fused silica WCOT capillary column  
coated with SE-54

Temperature Program: (A) 50°C for 4 minutes, then programmed to 260°C  
at 10°/minute  
(B) 70°C for 4 minutes, then programmed to 280°C  
at 8°/minute  
(C) 70°C for 4 minutes, then programmed to 300°C  
at 8°/minute

Injector Temperature: 280°C

Carrier Gas: Helium

Flow Rate: 20 ml/minute (A)

Injector: Modified German-Horning (B) and (C)

Injector Flow: 20 ml/minute (B) and (C)

Injector Pressure: 14 psi (B) and (C)

Split Ratio: Approximately 10:1 (B) and (C)

Separator: Glass jet (A)

Separator Oven Temperature: 280°C

Transfer Line Temperature: 280°C

MS Mode: Electron Impact (EI)

Electron Energy: 70 e/V

Electron Multiplier Voltage: Approximately 1700 V (without conversion  
dynode) or approximately 1000 V (with  
conversion dynode at 3000 V)

Emission Current: 0.2 ma

Preamplifier Sensitivity:  $10^{-7}$  a/v

Scan Range: 40-640 amu

Scan Time: (A) 3 seconds; (B) 2 seconds; (C) 1.5 seconds

Scan per Run: (A) 500; (B) 900; (C) 1400

Volume Injected: 2 or 3  $\mu$ l

Internal Standard: 100  $\mu$ g/ml  $d_{10}$ -anthracene in final extract

TALKS 4

All concentrations are in mM.

All concurrent transactions are in  $\text{WS}/\text{I}$  (PWB)

All coincident rat tones are in  $\log(1)$  (soft)

## All concurrent face functions are $\text{LR}$ wrt / (rank)

All subsequent hours are  $[n \cdot m]/(m+n)$

All concentrations are in  $\text{mg/l}$  (ppb)

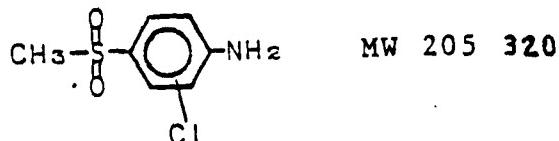
All concentrations are in  $\mu\text{g/l}$  (ppb)

Footnotes to Table 4

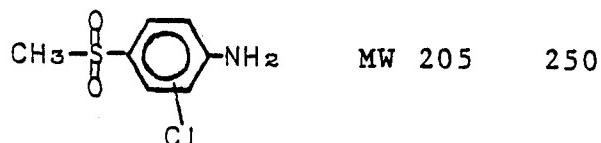
- 1) At least one ion in spectrum saturated; actual concentration probably greater.
- 2) Dimethyldisulfide 3600<sup>1)</sup>
- 3) 1,2-bis(methylthio)ethane 560
- 4) Dichloroethene 280
- 5) Cyclopentene 1900
- 6) Cyclopentadiene 400
- 7) Methyl cyclohexane 500
- 8) Dimethyldisulfide 4200<sup>1)</sup>
- 9) 1,2-bis(methylthio)ethane 590
- 10) Methyl cyclohexane 250
- 11) Ethyl benzene 4500
- 12) Hexachloroethane 1800
- 13) Dichloropropene 210
- 14) Bromochloropropene 310
- 15) Ethyl benzene 630
- 16) Bromodichloro methane 520
- 17) 2-propanol 360
- 18) Dichloroethene 300
- 19) Methyl cyclohexane 300
- 20) Ethyl benzene 340
- 21) Methoxy benzene 440
- 22) Dimethyl disulfido 1200<sup>1)</sup>
- 23) Cyclopentene 360
- 24) Cyclohexane 260
- 25) Hexane 400
- 26) Methyl cyclohexane 450
- 27) Ethyl benzene 490
- 28) Methyl cyclopentane 430
- 29) Dichloropropane 3800
- 30) Hexane 260
- 31) Methyl cyclohexane 420
- 32) Trichloropropene 370
- 33) Ethyl benzene 1900
- 34) Thiophene 200
- 35) 2-butanone 250
- 36) 2-butanone 440
- 37) Bromodichloromethane 1300
- 38) Bromochloropropene 1200
- 39) Bromochloropropane 320
- 40) Trichloroethanol 670
- 41) Methyl cyclohexane 200
- 42) Cyclopentene 380
- 43) Methyl cyclohexane 220
- 44) Cyclopentene 270
- 45) Methyl cyclohexane 490
- 46) Hexane 240
- 47) Methyl cyclohexane 390
- 48) Thiophene 320
- 49) Dichloropropane 380
- 50) Ethyl benzene 200
- 51) Ethyl benzene 440

All concentrations are in  $\mu\text{g/l}$  (ppb)

- 52) Dipropyl ether 250
- 53) Thiophene 240
- 54) Ethyl benzene 210
- 55) dipropyl ether 360
- 56) Methyl cyclopentane 590
- 57) Methyl cyclohexane 540
- 58) Ethyl benzene 730
- 59) Bromodichloromethane 8700<sup>1)</sup>
- 60) Dibromochloromethane 450
- 61) Ethyl benzene 400
- 62)



- 63) Dibenzyl ether 350
- 64) From packed column BN run
- 65) From capillary column BA run
- 66) Ethyl benzene 240
- 67) Ethyl benzene 1000
- 68) Dichloroethylene 970
- 69) Methyl cyclohexane 580
- 70) Carbon disulfide 360
- 71) Cyclopentene 230
- 72) Cyclohexane 210
- 73) Thiophene 550
- 74) Methyl cyclohexane 610
- 75) Ethyl benzene 310
- 76) Diphenyl ether 380
- 77) VOA sample taken from quart jar
- 78)



- 79) Cyclopentene 360
- 80) Cyclohexane 260
- 81) Hexane 400
- 82) Methyl cyclohexane 450
- 83) Ethyl benzene 490
- 84) Dipropyl ether 320
- 85) Sample lost
- 86) RIR-24-642-79
- 87) RIR-24-643-79
- 88) Average of two determinations
- 89) Trichlorophenol 910

All concentrations are in  $\mu\text{g/l}$  (ppb)

Bromochloromethane

Internal Standard

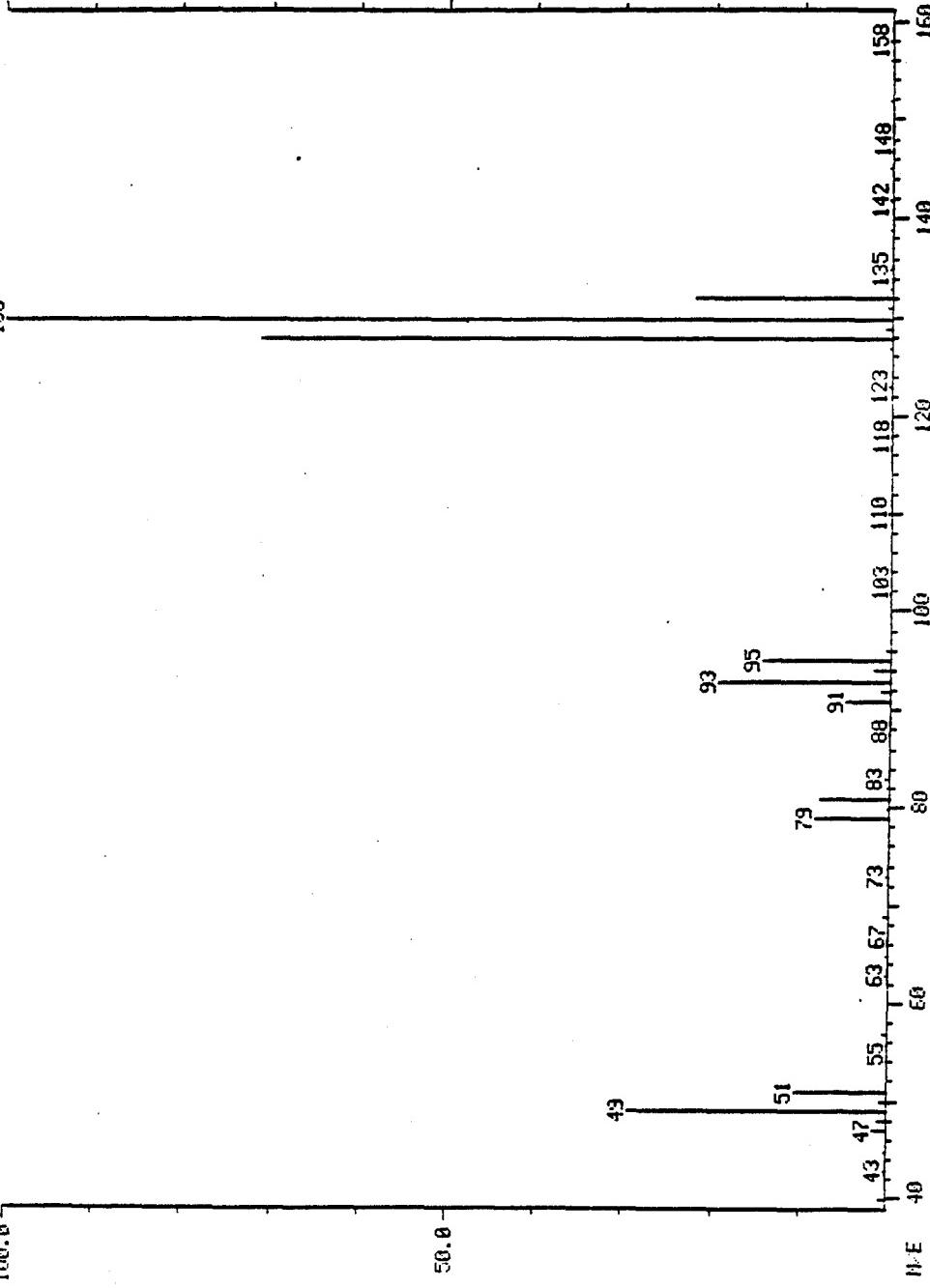
3EE88.

133

166.6

50.0

R.E



206832.

Chloroform

83

85

87

47

93

97

109

120

123

128

135

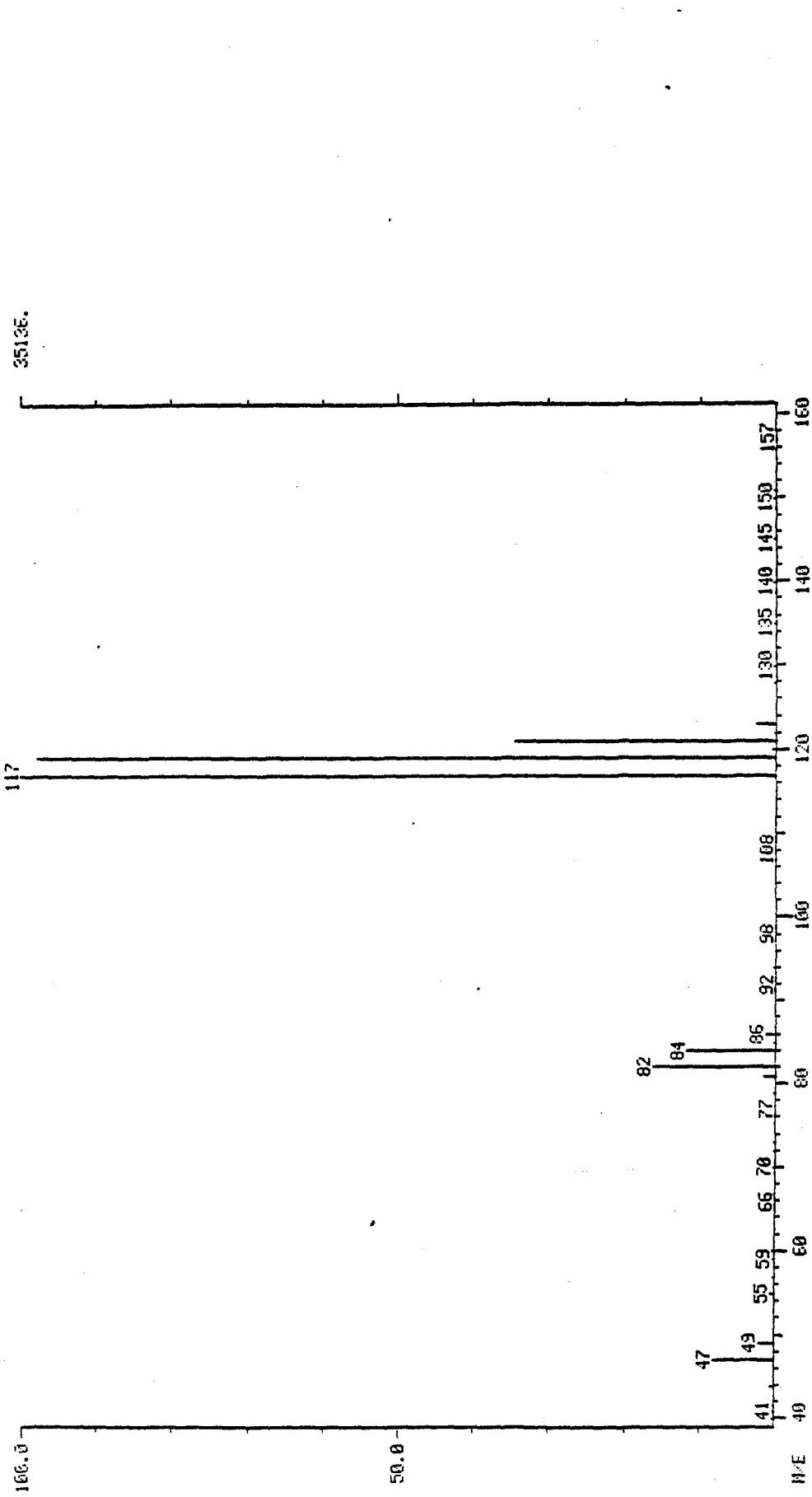
140

58.0

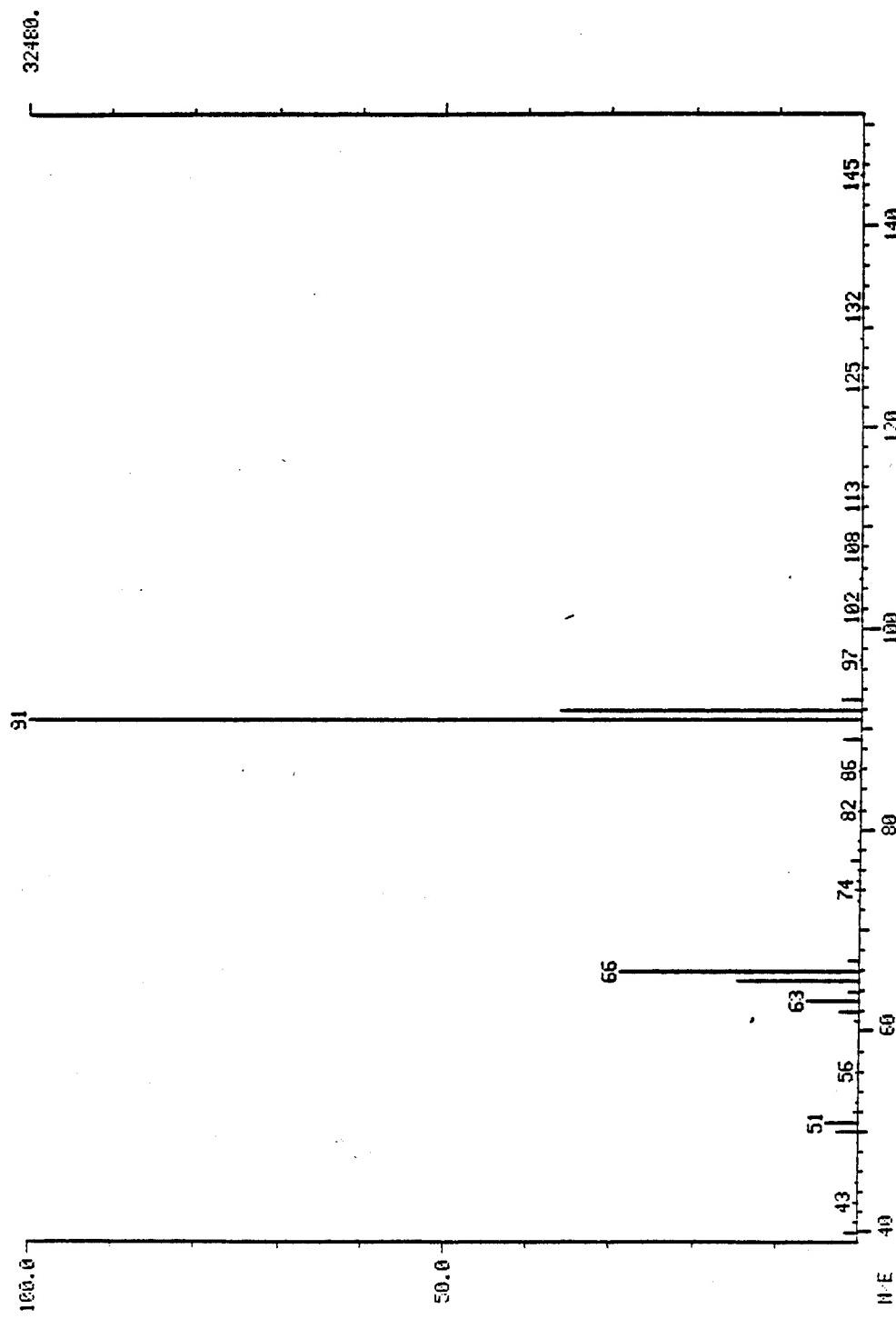
100.0

4.4

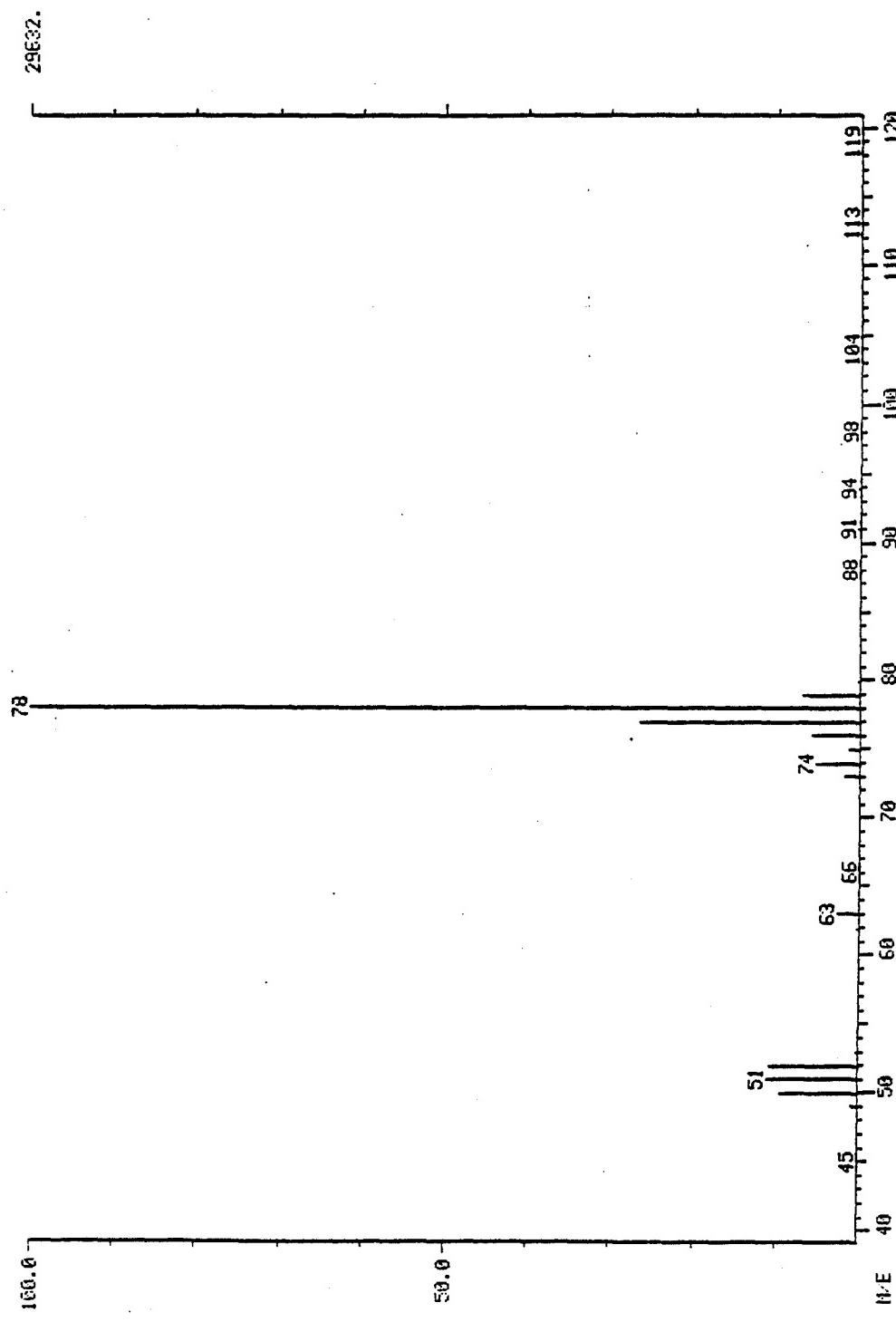
Carbon Tetrachloride

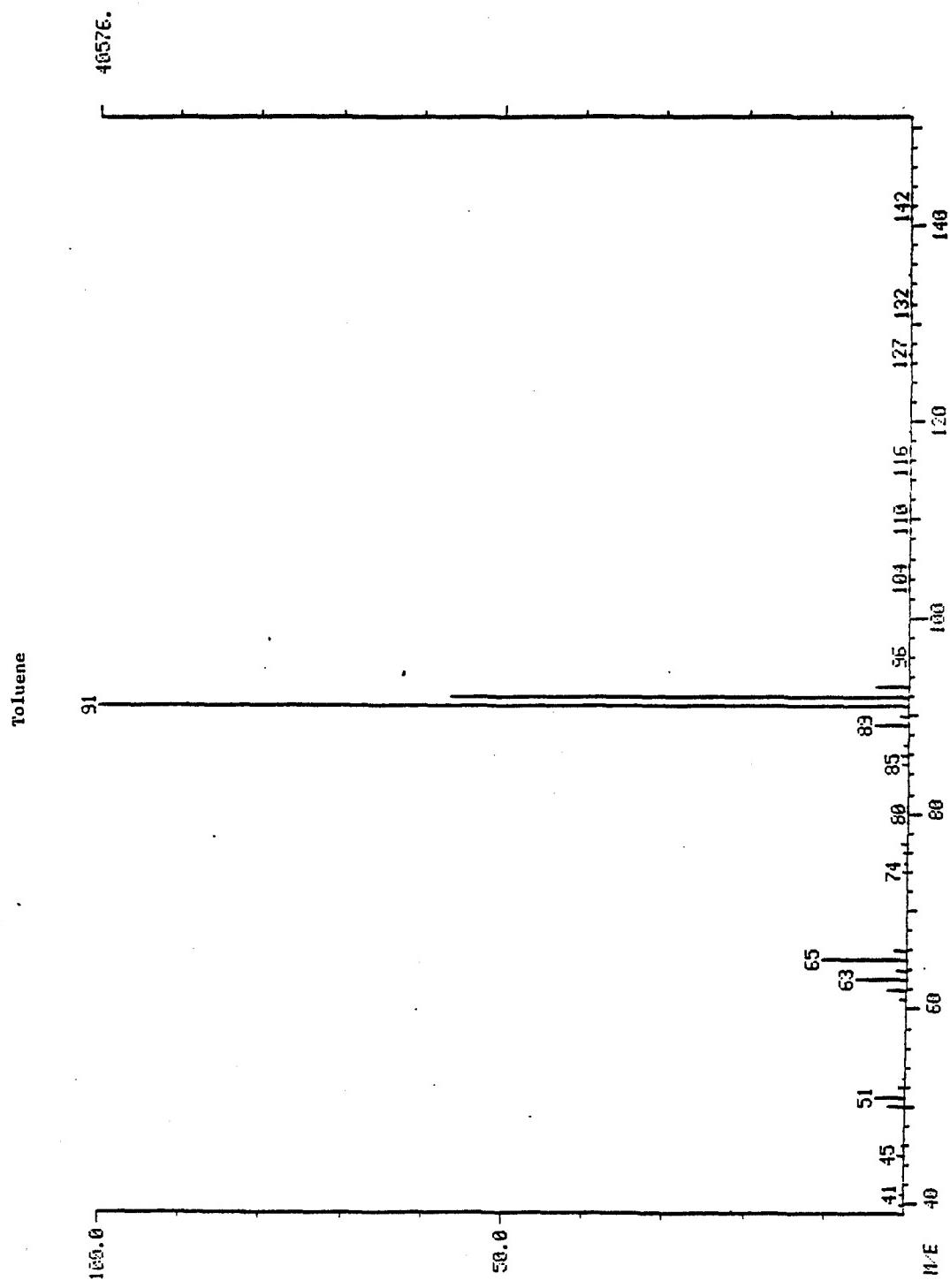


Bicycloheptadiene

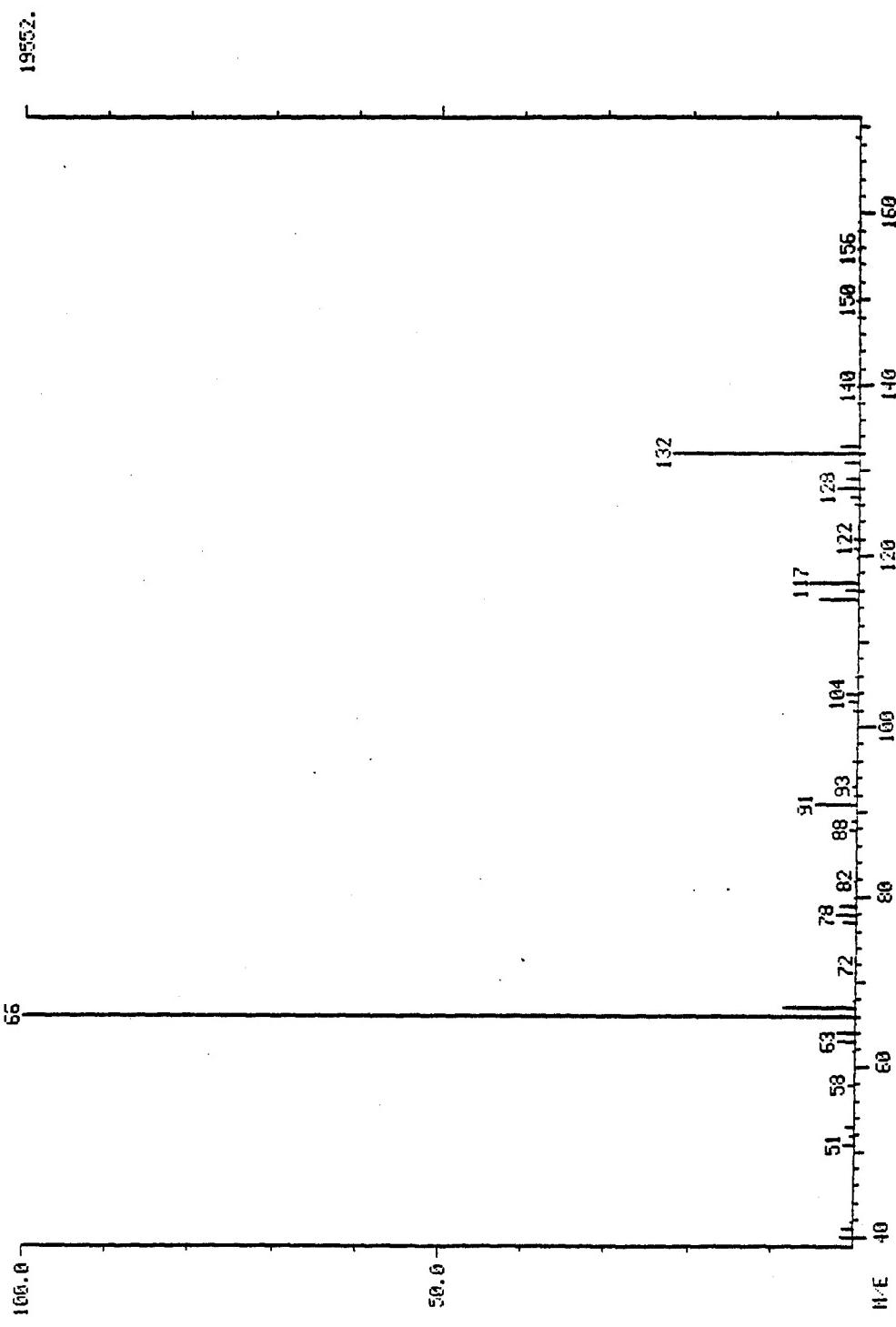


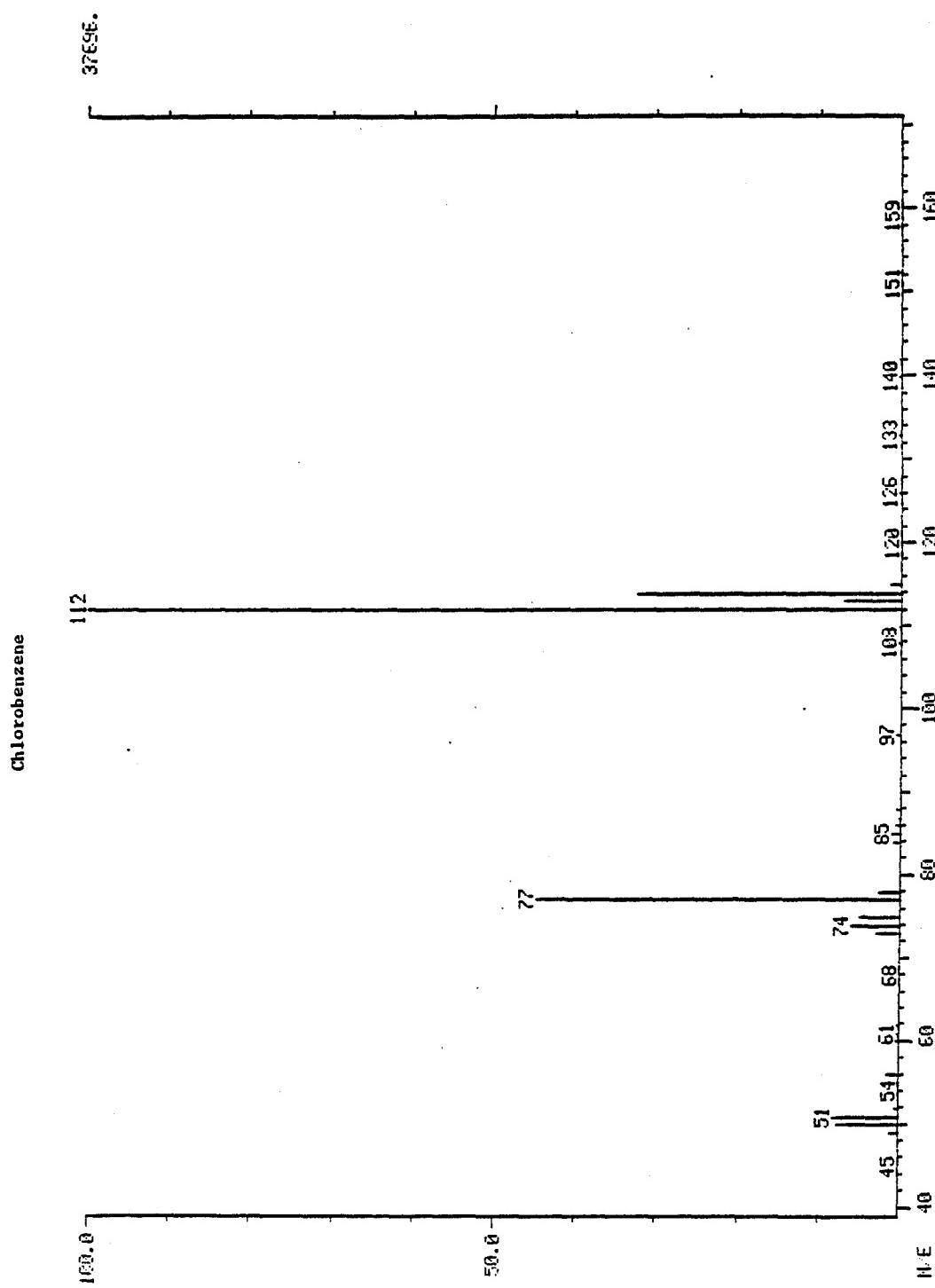
Benzene



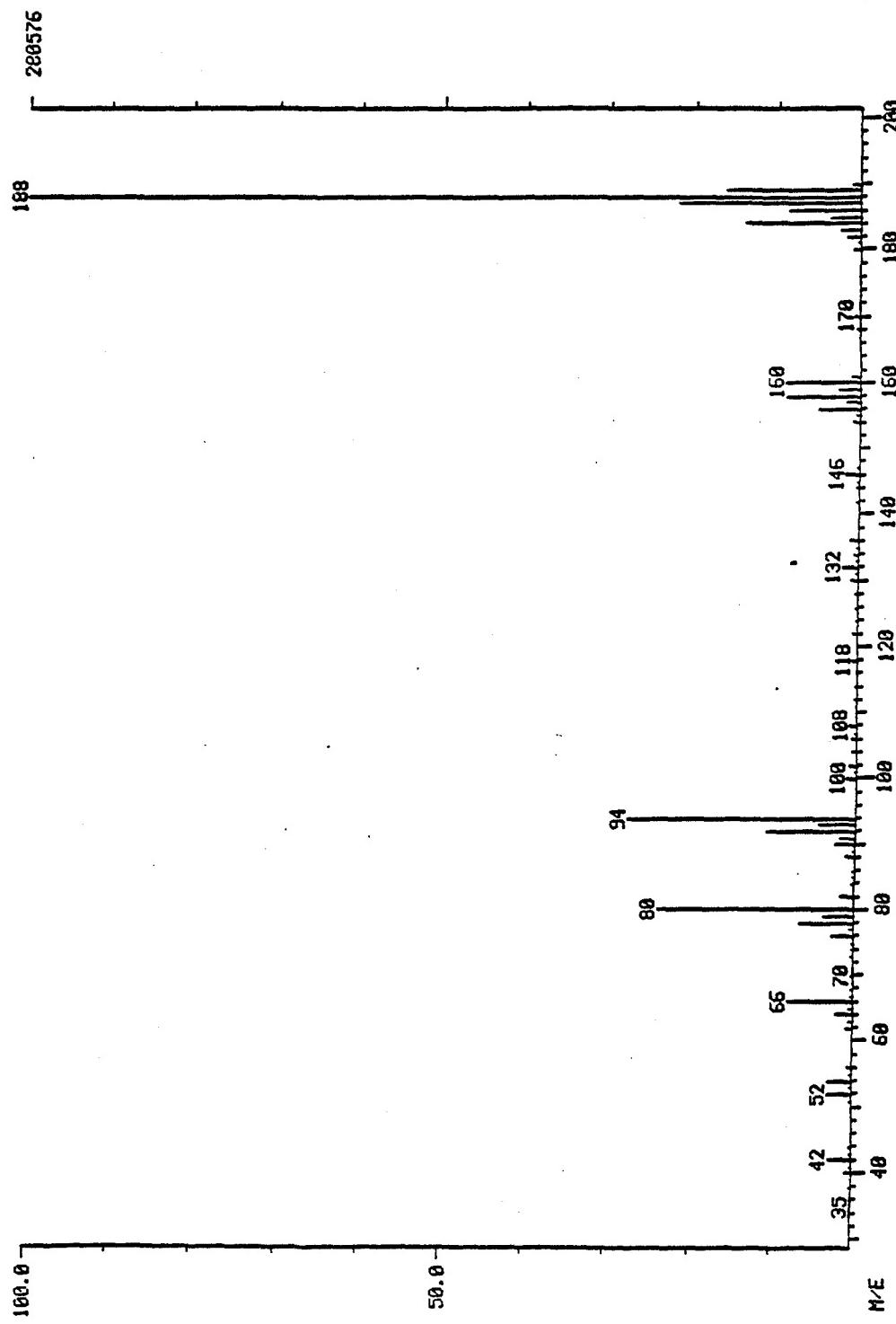


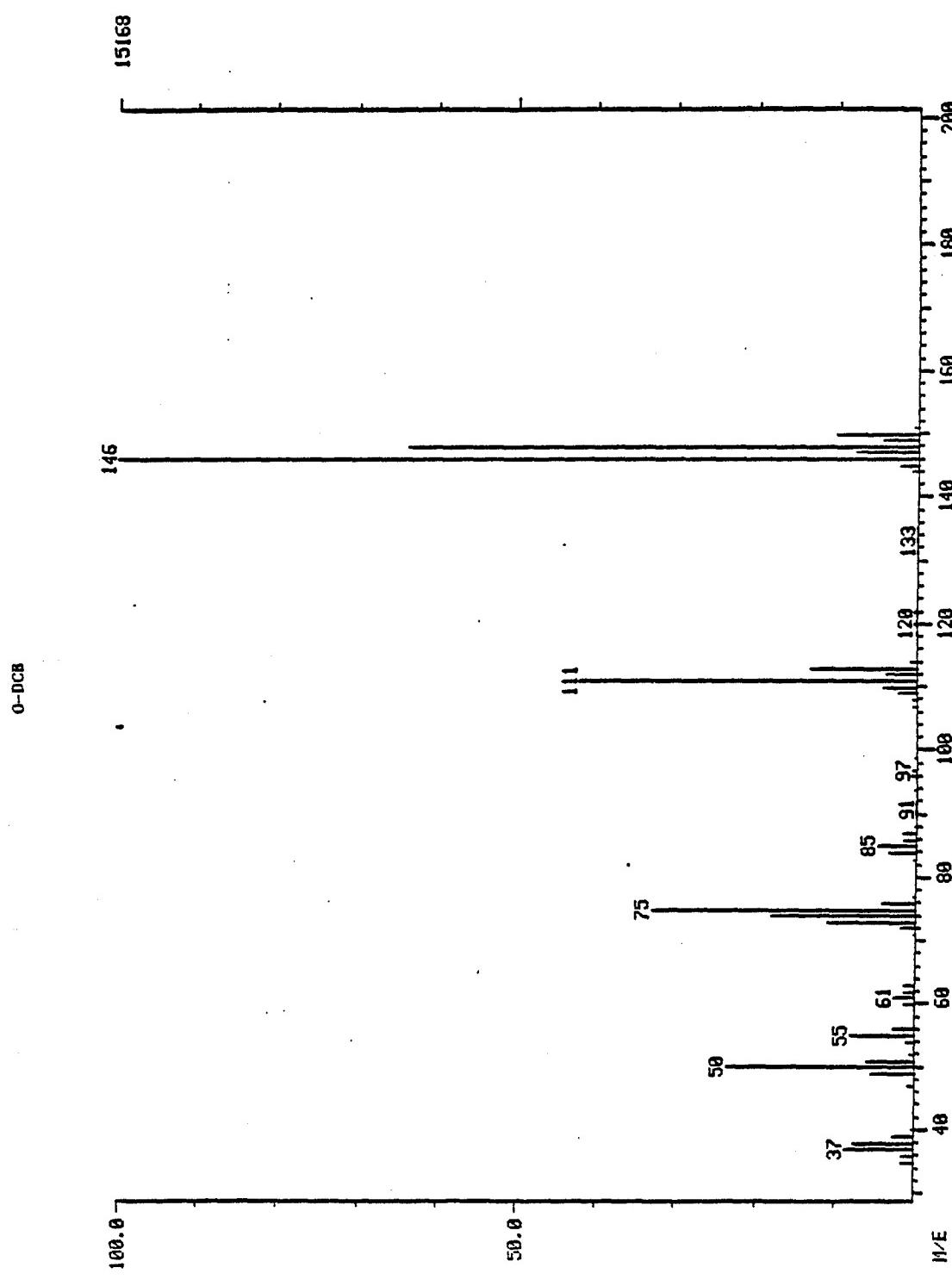
Dicyclopentadiene



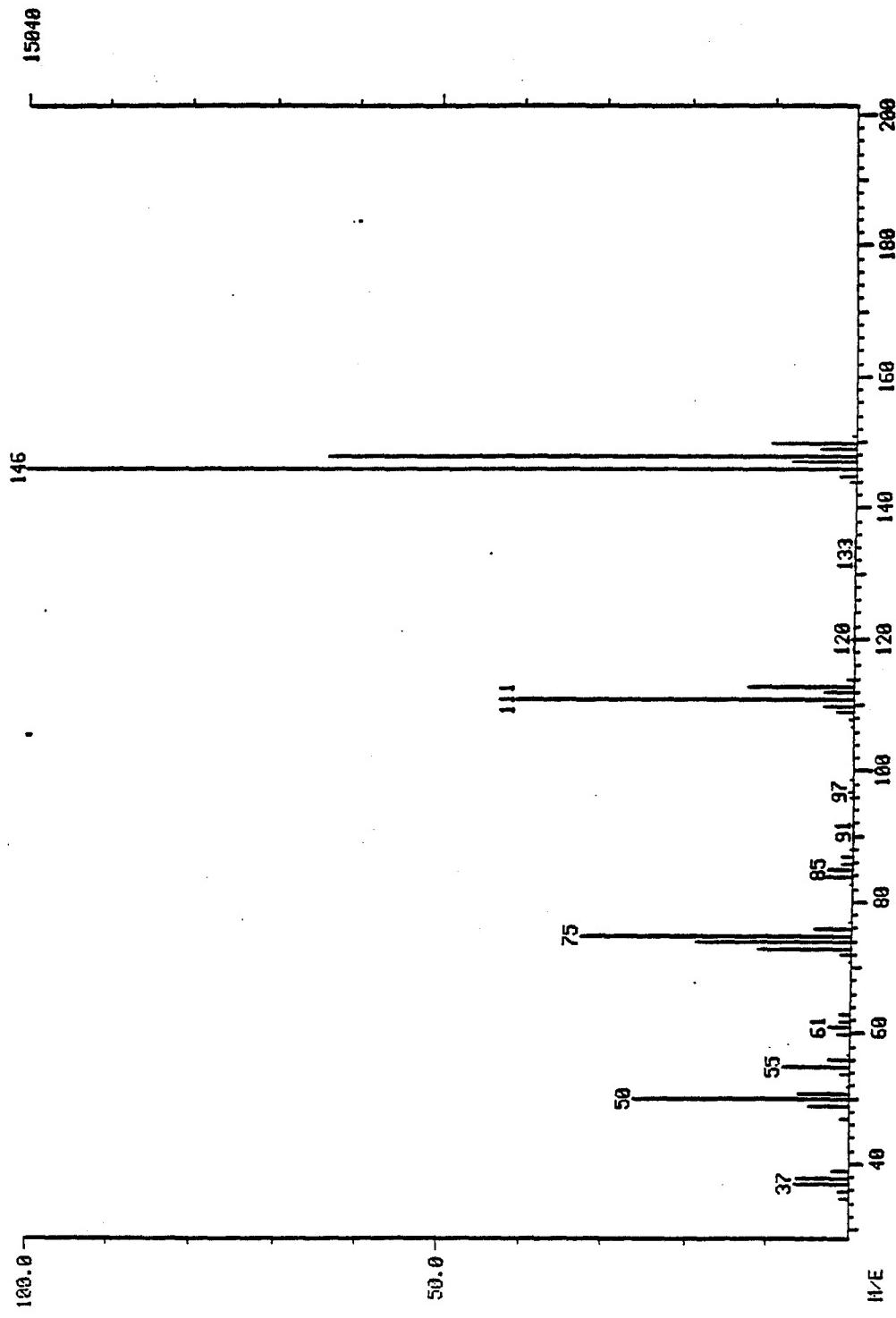


Dio-Anthracene

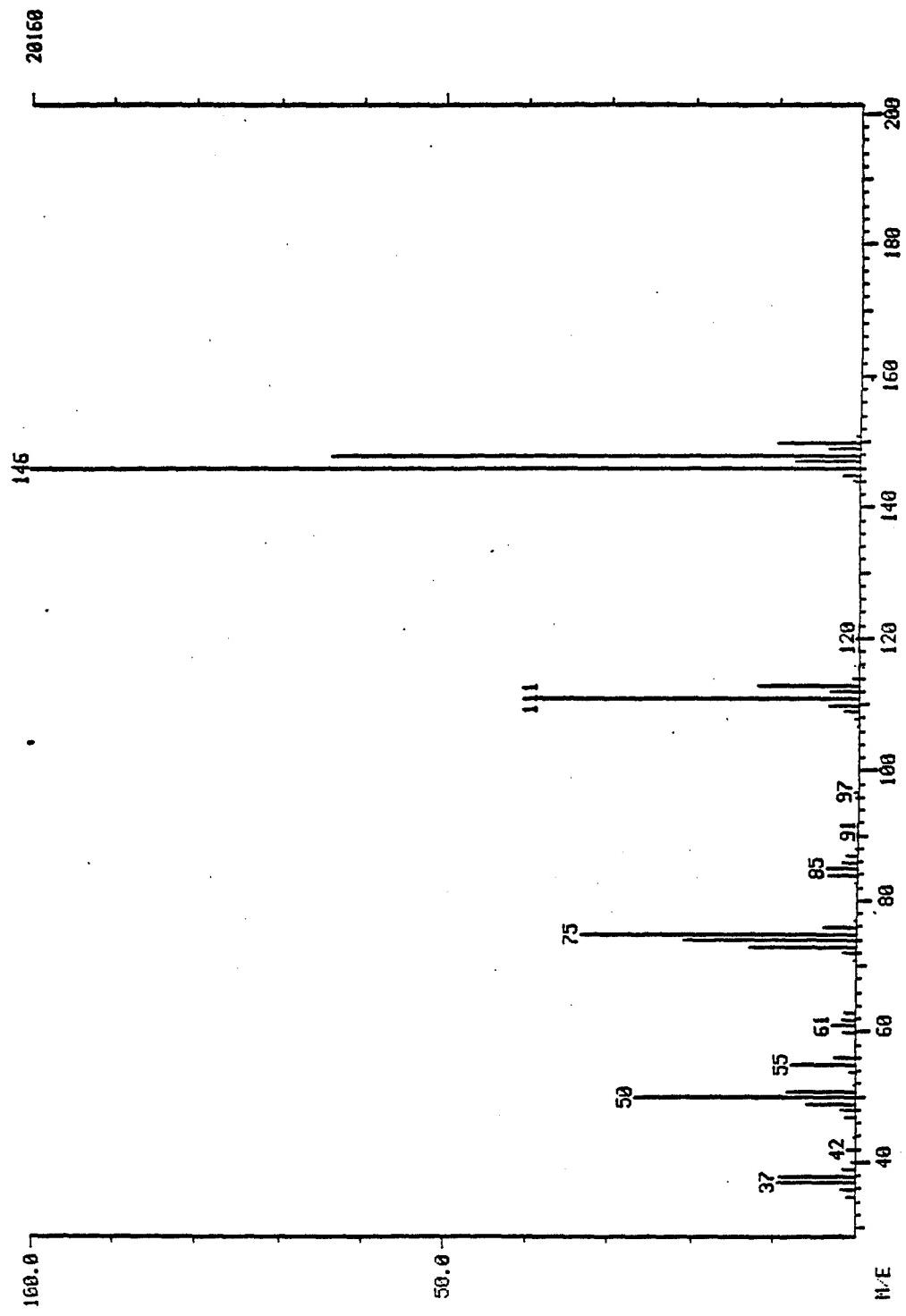


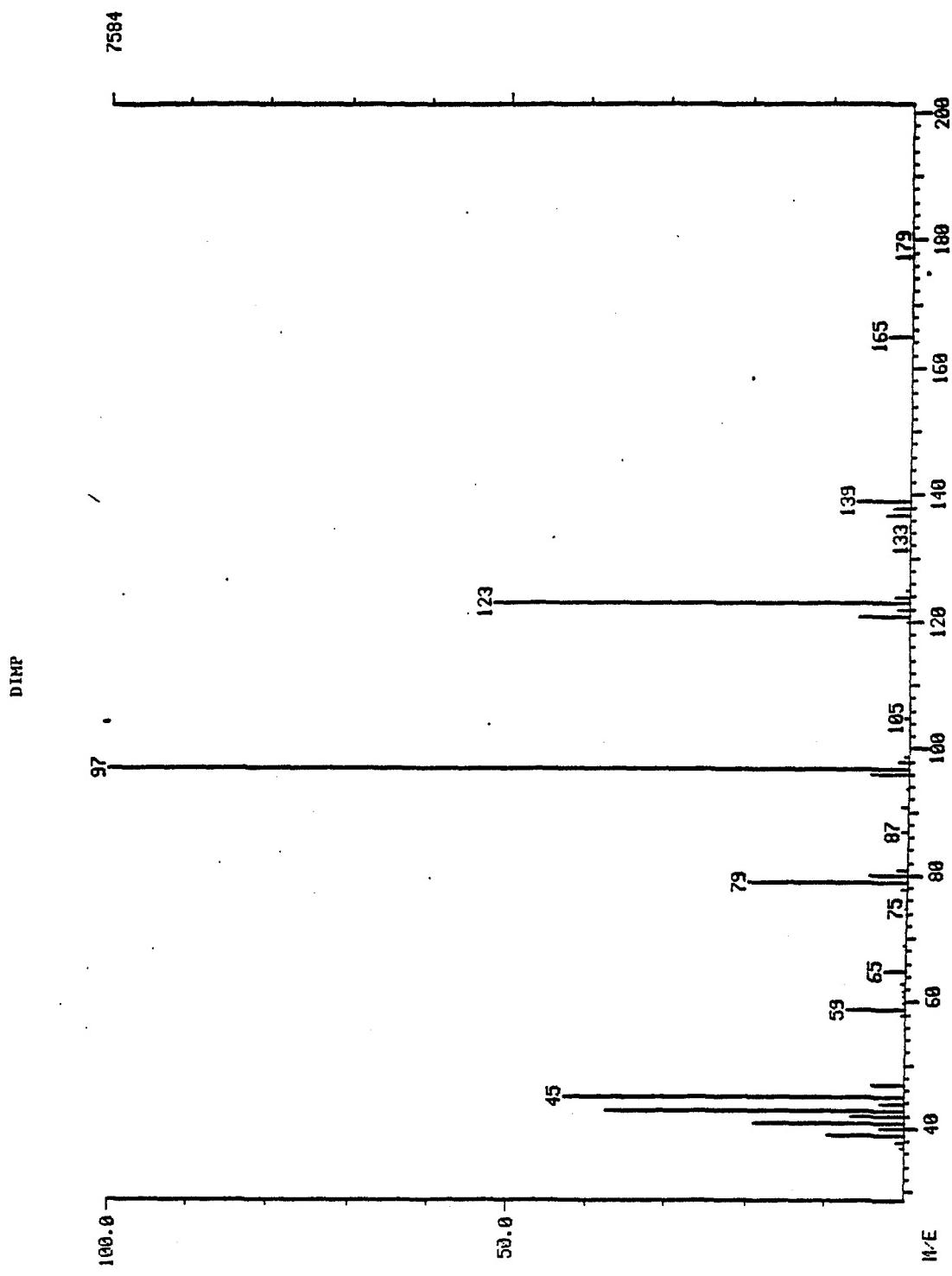


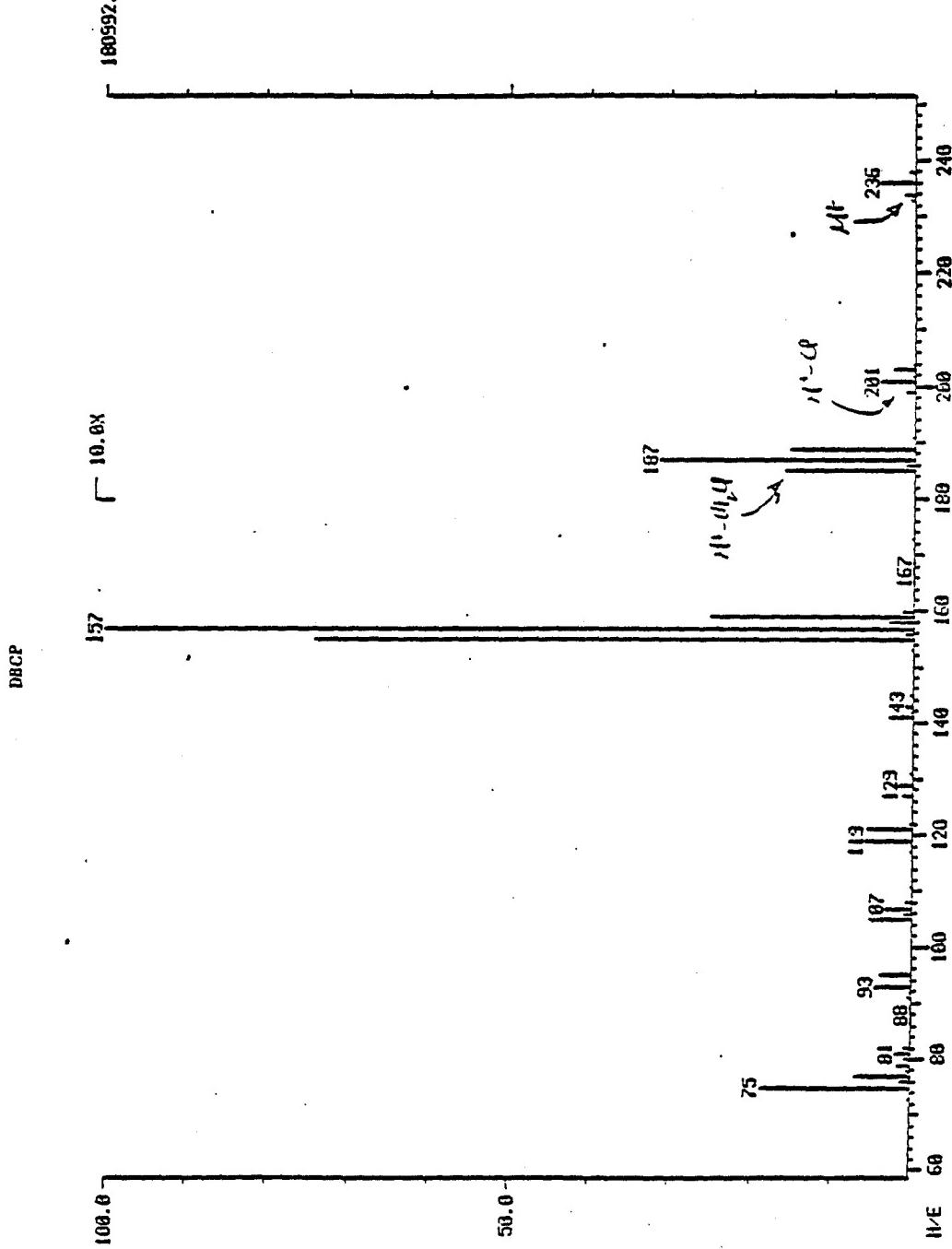
H-DCB



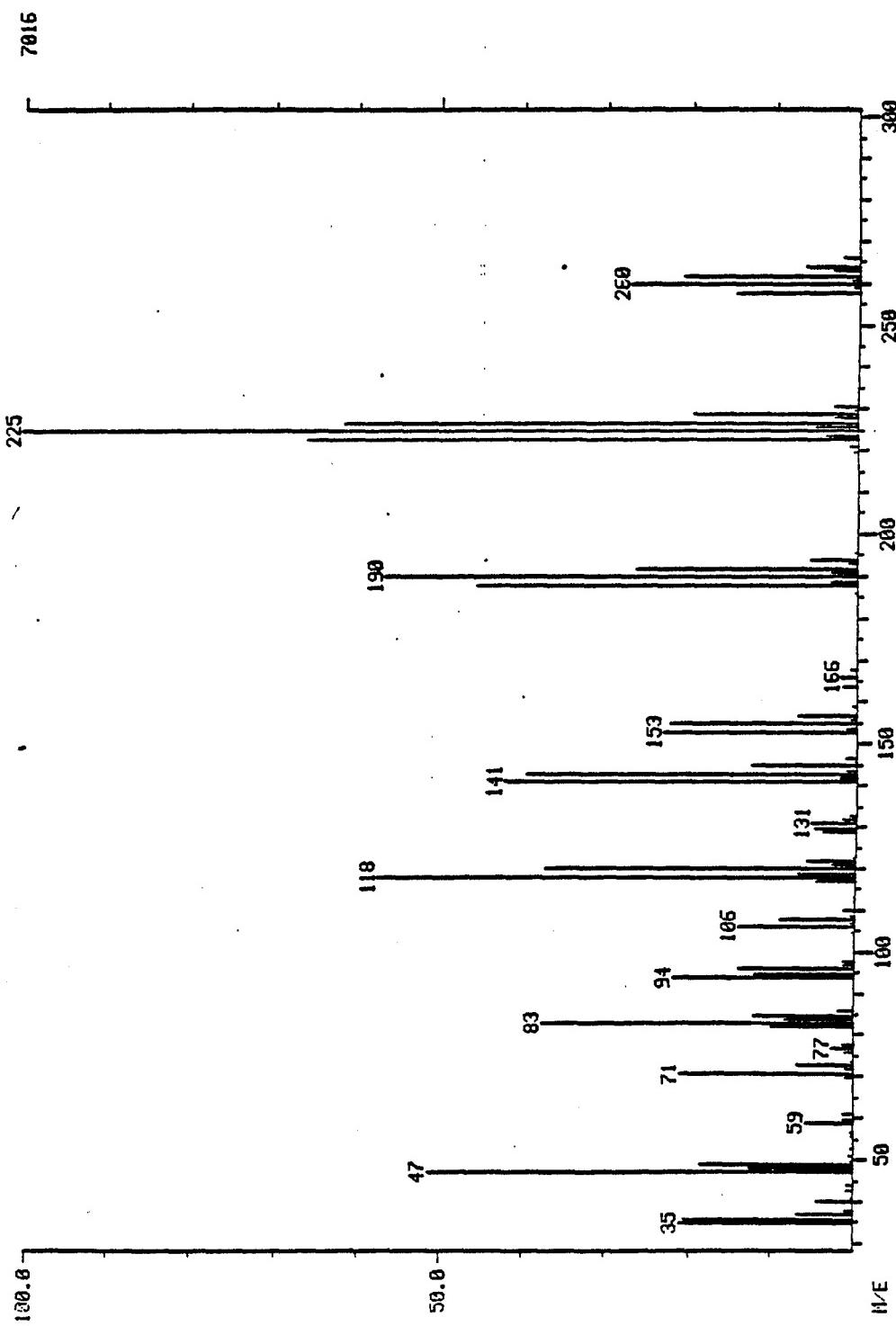
P-DCB



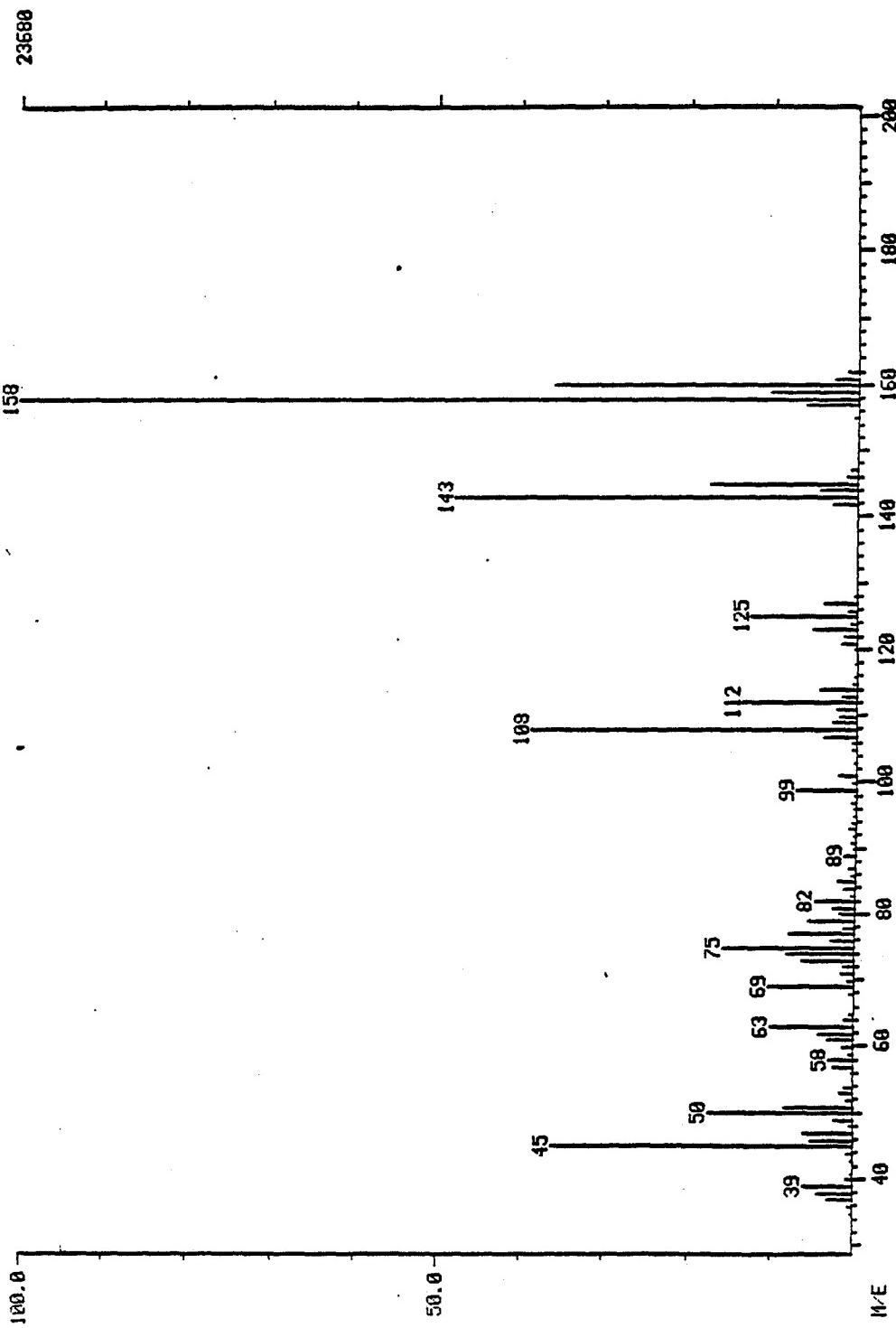




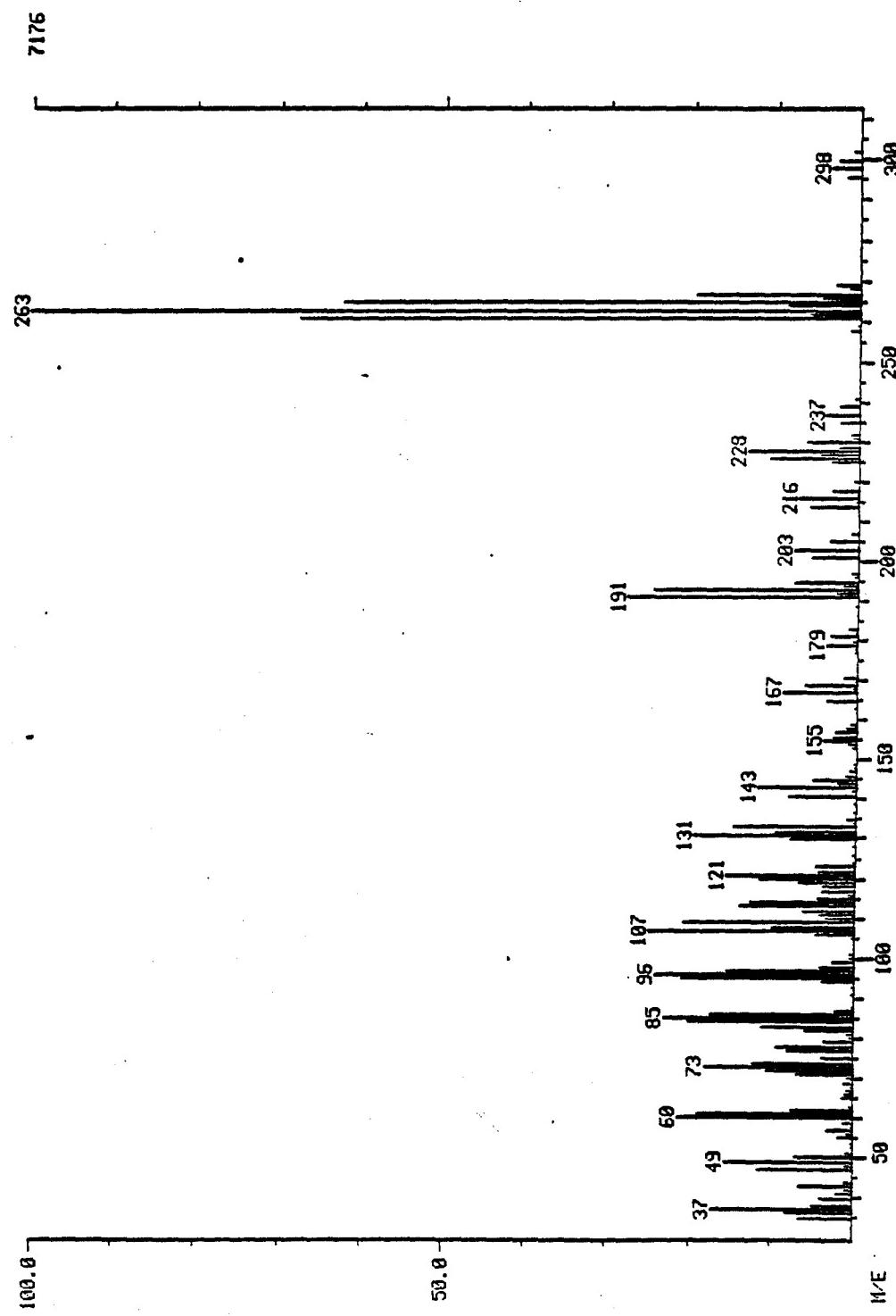
HCBD



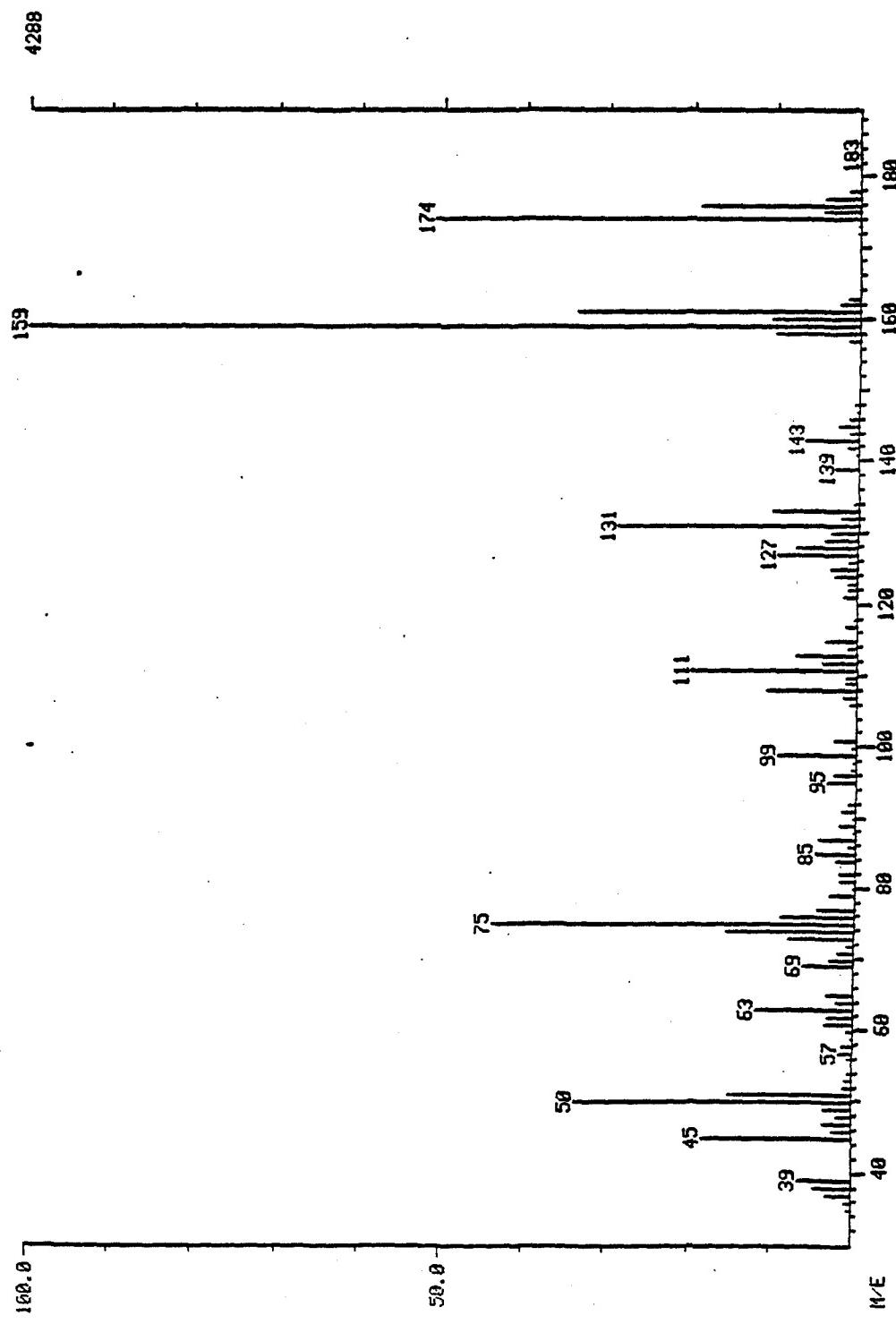
SD 9636

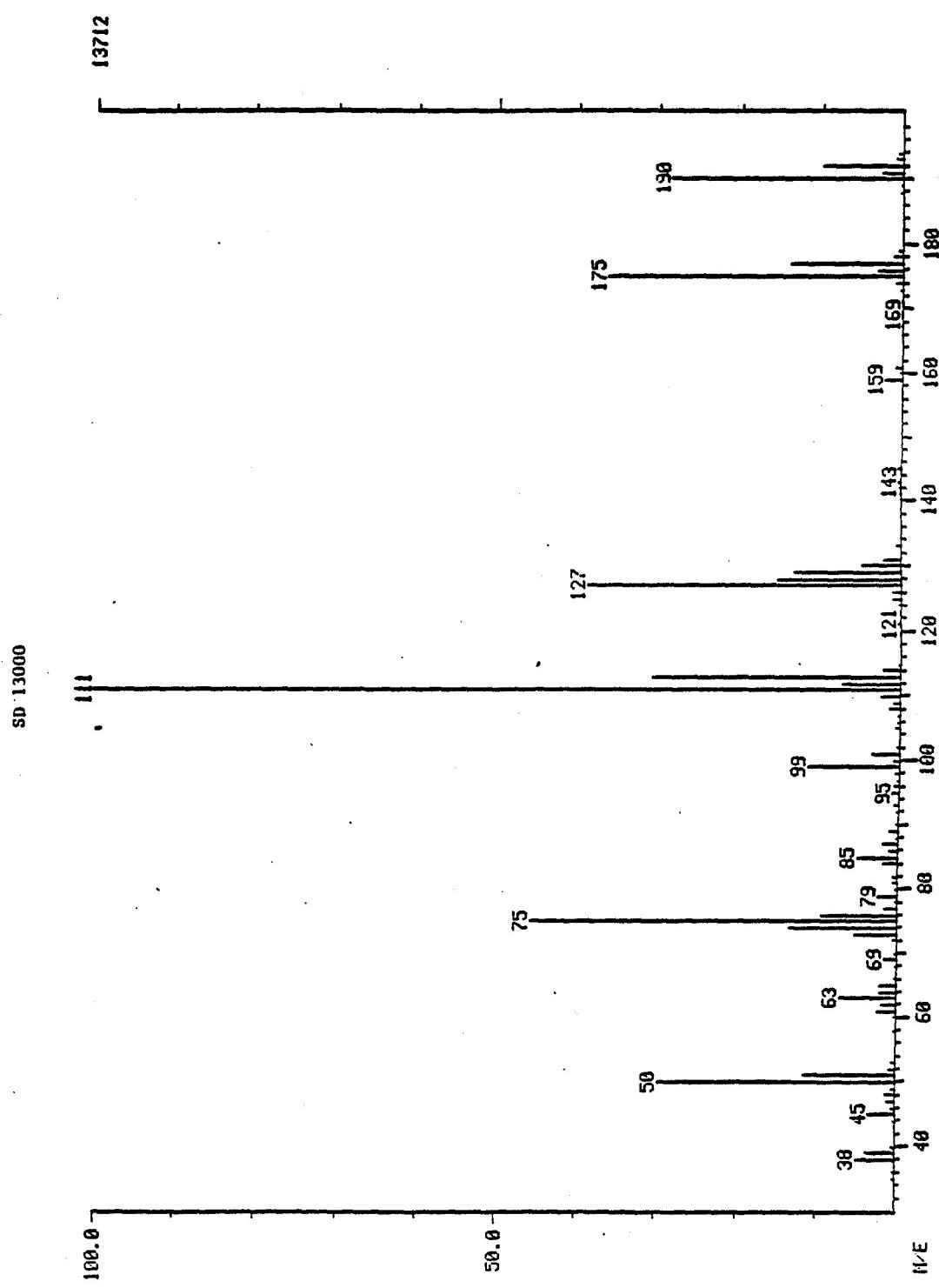


HCNB

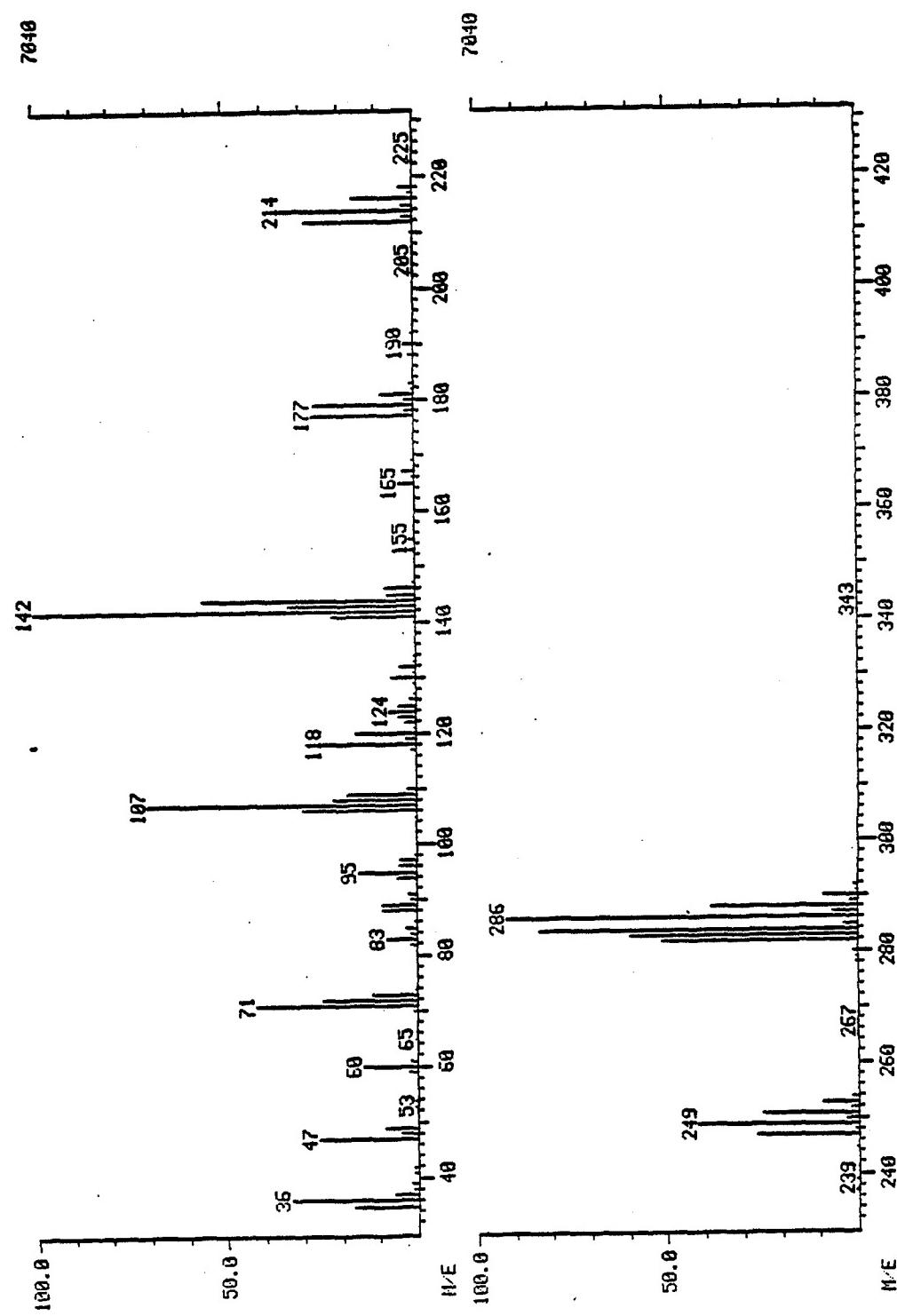


SD 13957

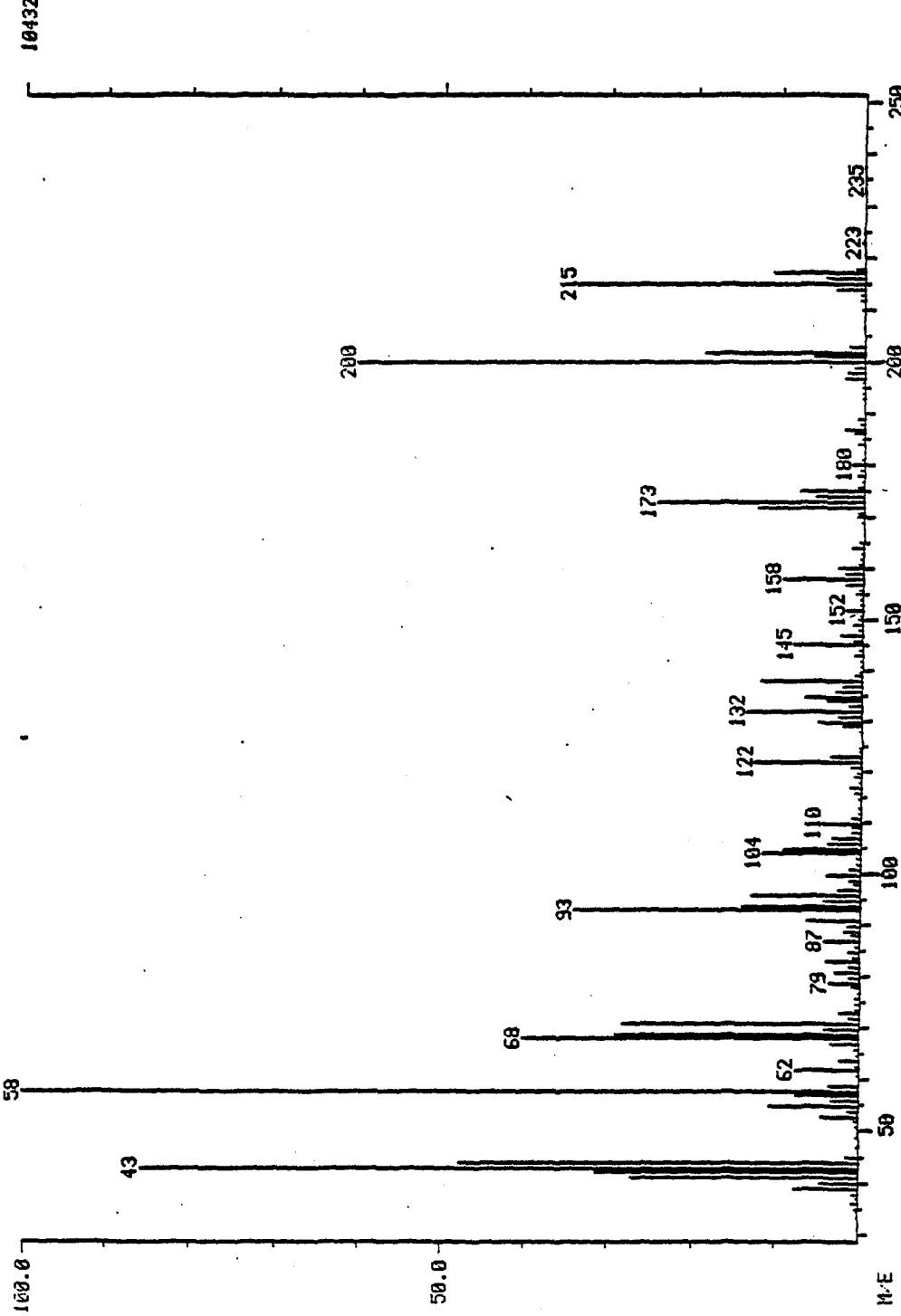




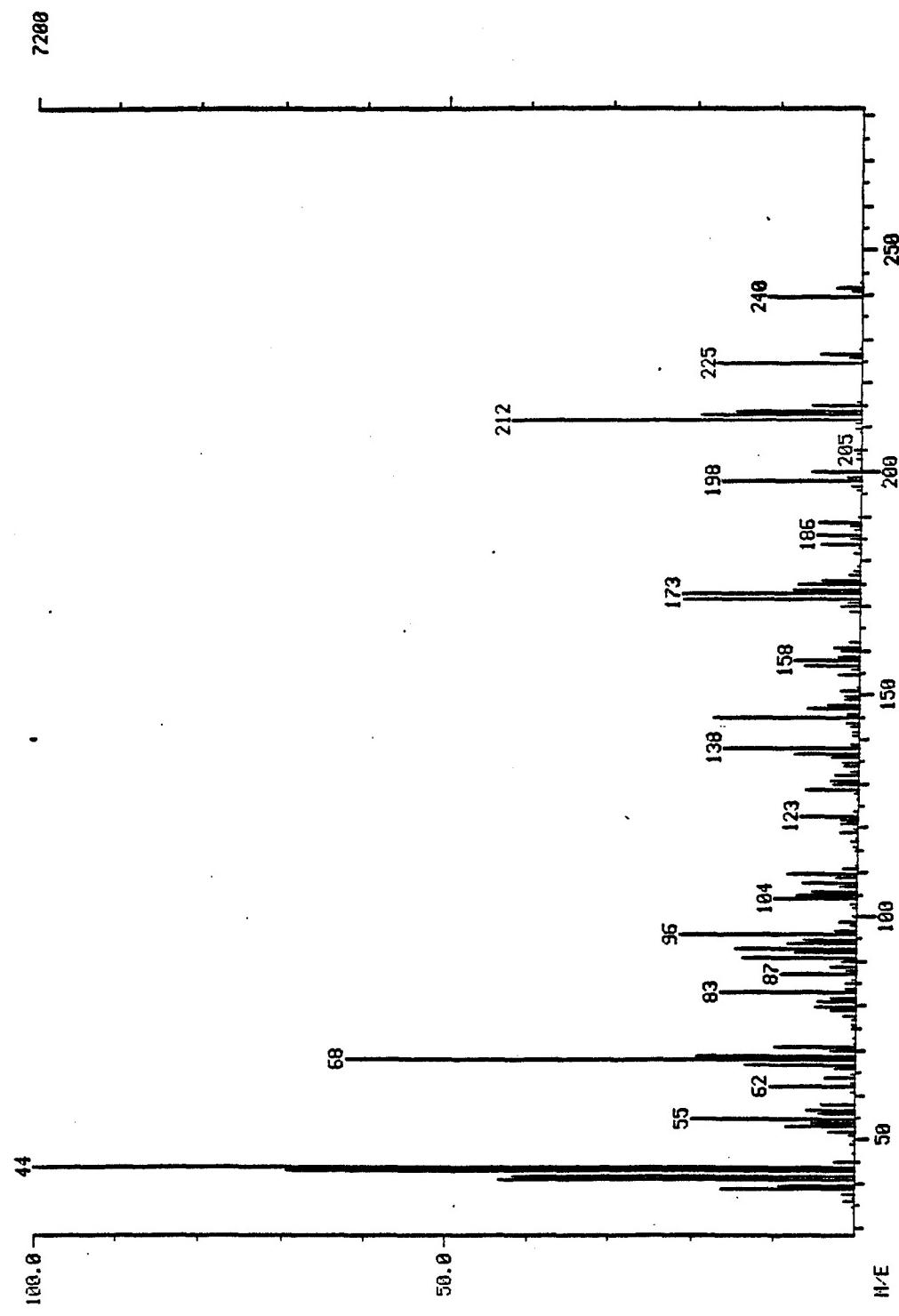
HCR



Atrazine



BLADEX



HEAD

,  
10.6X

32320

100.0

50.0

H/E

10.0

79

82

53

57

55

85

97

108

109

113

121

133

143

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170

~~2800~~ 3000 4500?

W 200-2

• 18a 5'

W 1000'

w 500'

**SECTION 35**

SECTION - 2

四

100

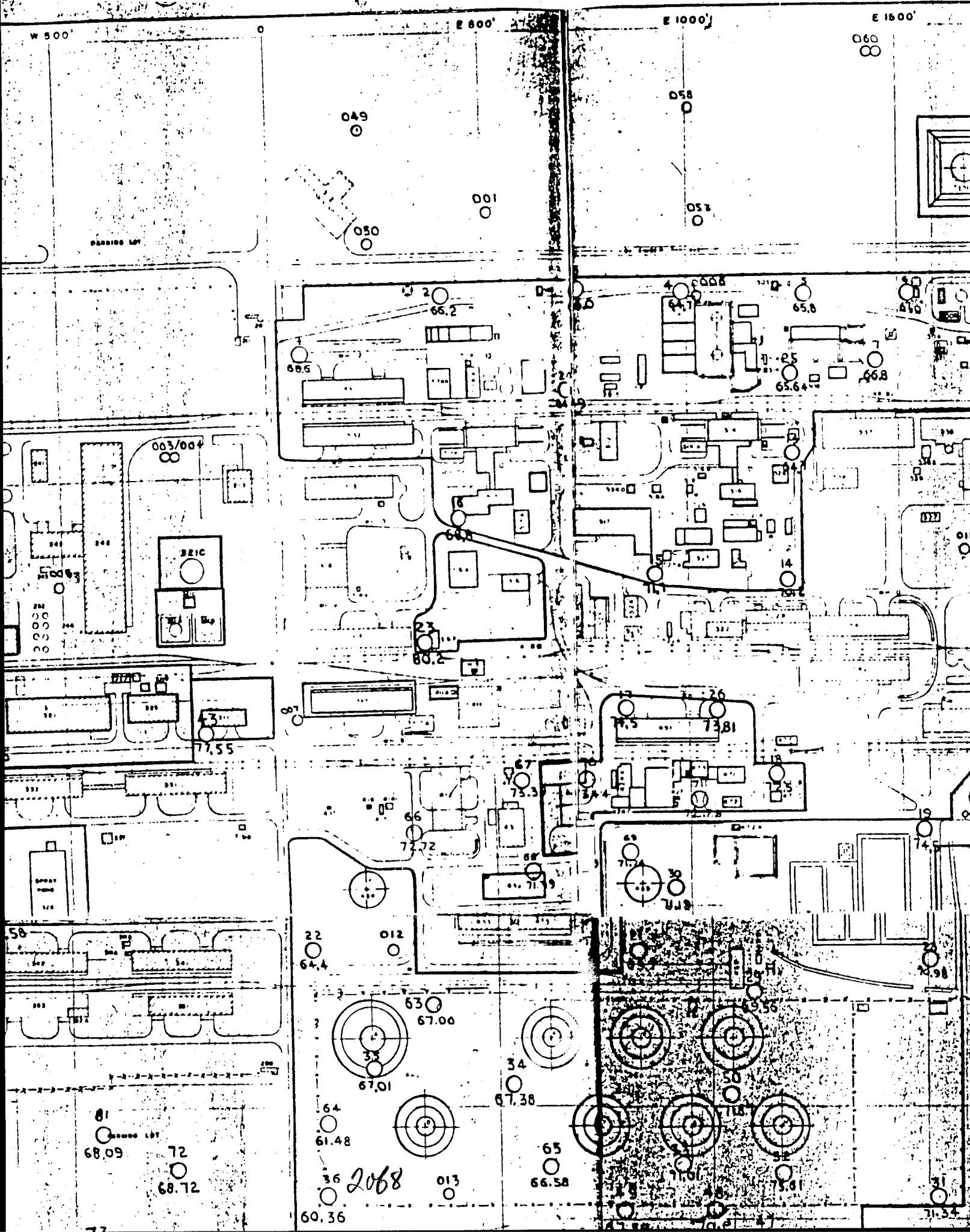
580

卷之三

-6-58

81

1068



E 15.00

E 2000

E 2500'

E 3000'

E 3800'

DWG. NO.	DESCRIPT
10002-1000	1000
10002-1010	1010
10002-1020	1020

REF. DWGS

០៧៦

2

## LEGENDI

~~2-5~~ HOLE

3068

### **Example**

66.2 -

60.2 ELEVATION TOP F

E 2600'

E 3000'

E 3600'

N 3000'

## DWG. NO.

## DESCRIPTION

REF. DWGS.

SECTION 36

SECTION 1

S 3000

S 3000

S 2000

## LEGEND

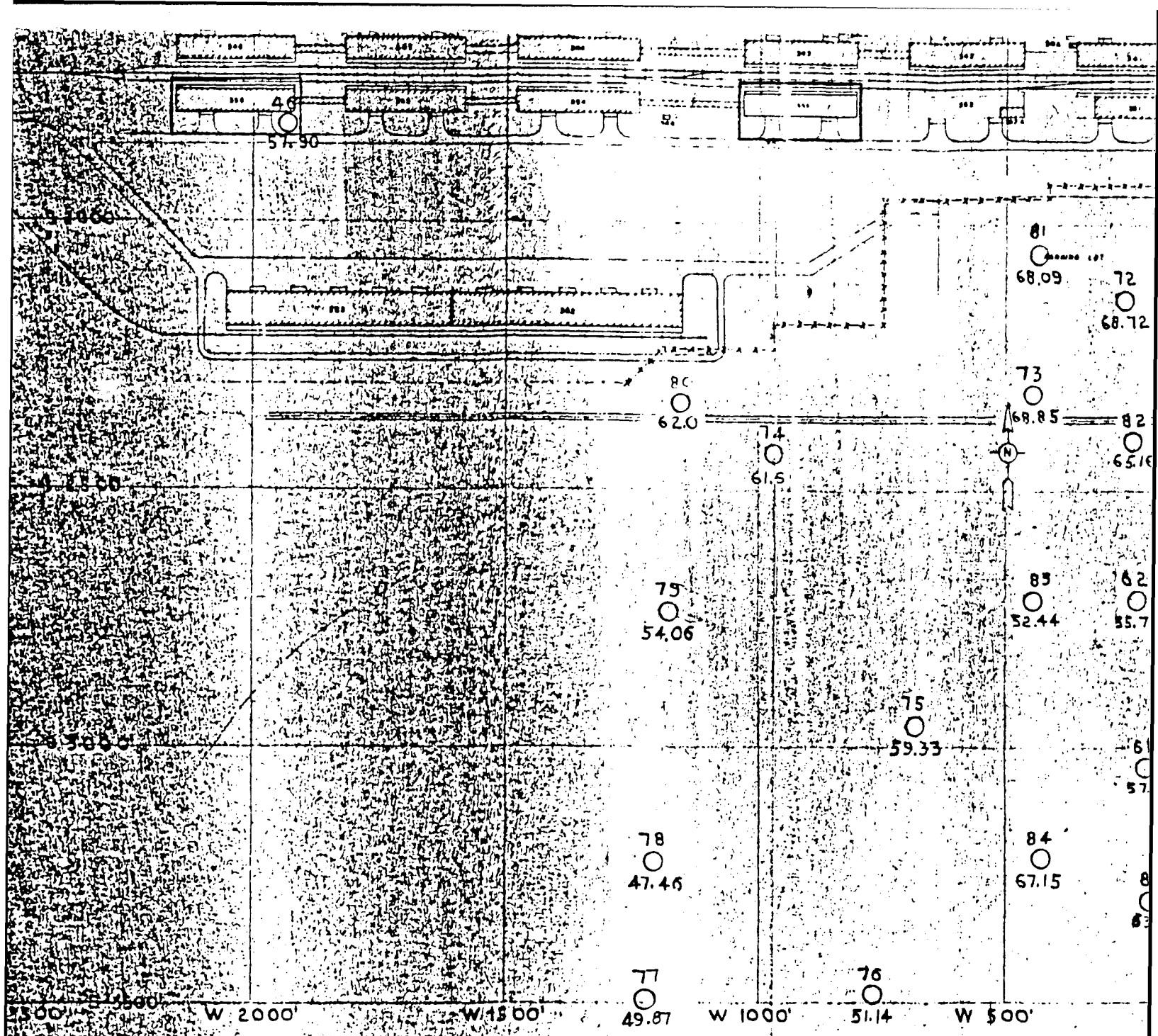
MOLE

EXAMPLE

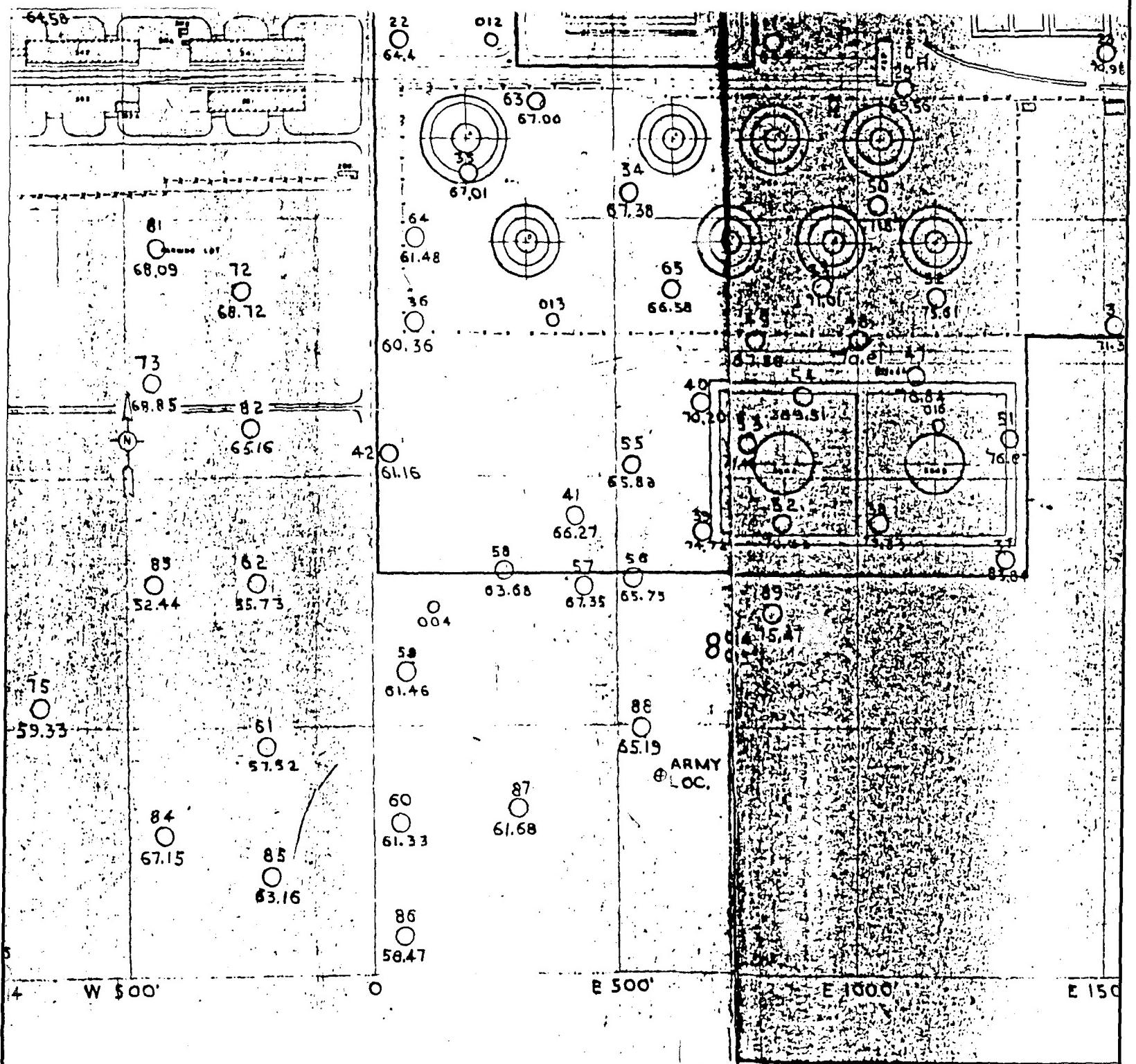
66.2

ELEVATION TOP PIPE (+5200')

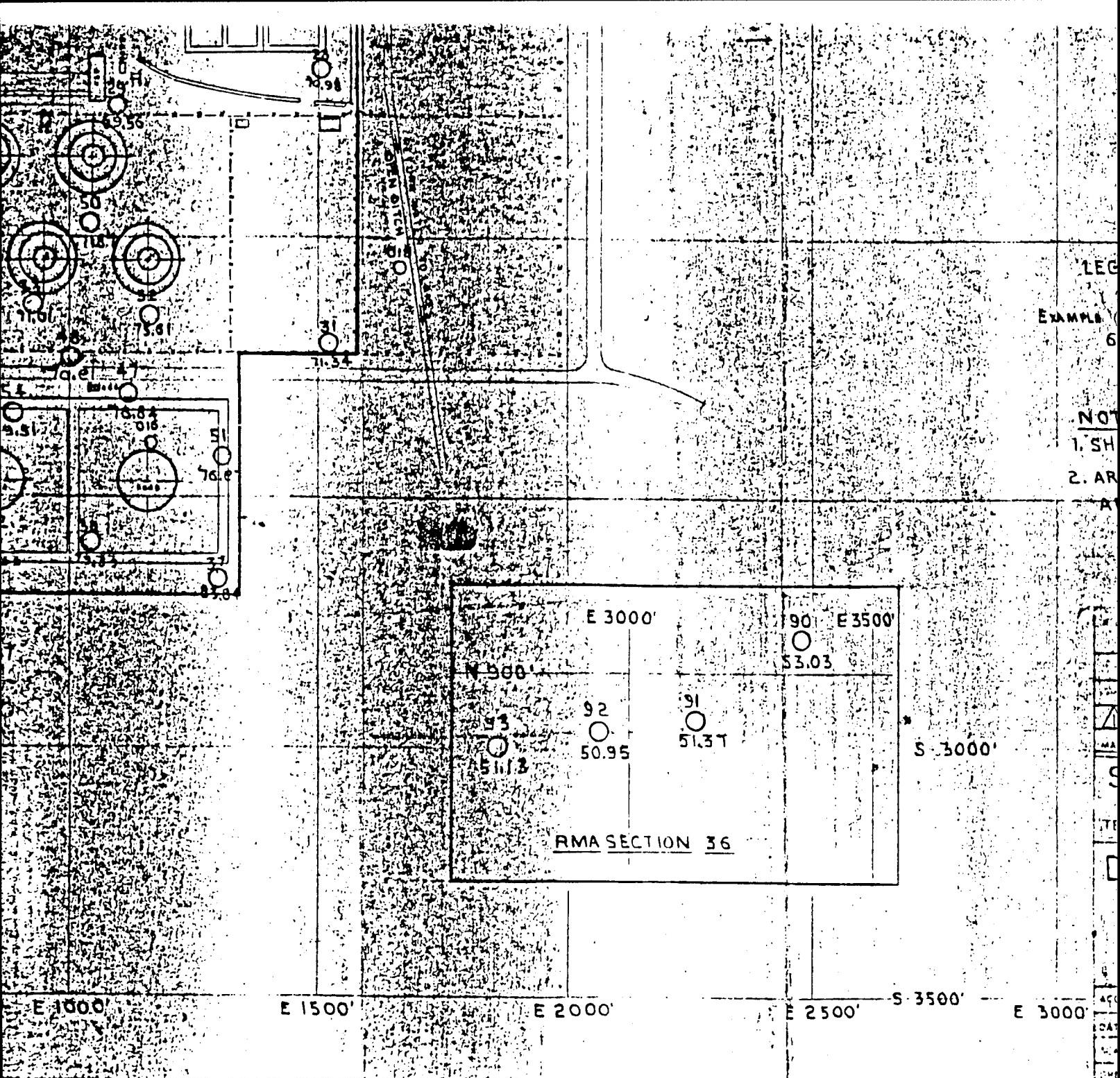
4068



5088



608



7088

S 2000'

LEGEND

EXAMPLE  
1 2 HOLE

662 ELEVATION TOP PIPE (+5200')

NOTES

1. SHELL OBSERVATION HOLES ARE  $\frac{3}{16}$ "

2. ARMY OBSERVATION HOLES

ARE  $\frac{1}{8}$ "

S 2500'

E 3000'

90 E 3500'

53.03

92

91

50.95

51.37

RMA SECTION 36

S 3000'

E 2000'

E 2500'

S 3500'

E 3000'

MARK	DATE	ACCT NO.	LIST OF REVISIONS
△ NIS 6872 UPDATE			
SHELL CHEMICAL COMPANY			
A DIVISION OF SHELL OIL COMPANY			
TECHNICAL DEPARTMENT		DENVER PLANT	
DENVER CHEMICAL PLANT			
83041R01			
2nd Copy			
ACCT NO 8042		PAGE 0000	
DATE 9-7-79		APPROVED WDS	
MAIL IN 200' DEN DLH		YE 13821-1	
CHECKED			

808